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EBOOK: Essentials of Economics, 10/e SCHILLER, 2016-03-22 EBOOK Essentials of Economics 10 e **Hands-On**

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**Praxisbuch Large Language Models** Jay Alamar, Maarten Grootendorst, 2025-10-28 Alle Werkzeuge und Techniken für die praktische Arbeit mit LLMs. Das Handbuch für das intuitive Verständnis von LLMs. Mit zahlreichen Visualisierungen, die Konzepte schnell zugänglich machen. Themen sind die Sprachverarbeitung, Textklassifikation, Suche oder Cluster und die Sprachgenerierung vom Prompt Engineering bis zur Retrieval Augmented Generation (RAG). Die Autoren haben mit ihren beliebten Blogs Millionen von Entwickler:innen geholfen, Machine Learning und KI zu verstehen. Diese umfassende und anschauliche Einführung in die Welt der LLMs beschreibt sowohl konzeptionelle Grundlagen als auch konkrete Anwendungen und nützliche Tools. Tauchen Sie in das Innenleben von LLMs ein und erkunden Sie ihre Architekturen, Einsatzbereiche, Trainingsmethoden und Feintuning-Techniken. Mit seiner einzigartigen Mischung aus intuitiv verständlichen Illustrationen und praxisbezogenen Erläuterungen ist dieses Buch die ideale Ausgangsbasis für alle, die die Möglichkeiten von KI-Systemen voll ausschöpfen möchten. Sie lernen vortrainierte Transformer-LLMs von Hugging Face für Anwendungsfälle wie das Verfassen von Texten oder für Inhaltssammenfassungen einzusetzen. Sie erfahren außerdem, wie Sie Suchsysteme erstellen und vorhandene Bibliotheken und vortrainierte Modelle für Textklassifikation, Suche und Clustering nutzen. Verstehen Sie die Architektur von Transformer-basierten Sprachmodellen, die bei der Textgenerierung und -repräsentation hervorragende Ergebnisse liefern. Entwerfen Sie fortgeschrittene LLM-Pipelines, um Textdokumente zu clustern und die darin enthaltenen Themen zu erforschen. Erstellen Sie semantische Suchmaschinen, die über den Abgleich von Schlagwörtern hinausgehen und auf Methoden wie Dense Retrieval und Reranking basieren. Lernen Sie, wie Sie generative

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web 33mol o<sub>2</sub> 6mol br<sub>2</sub> 11mol o<sub>2</sub> 18mol br<sub>2</sub> identify the limiting reactant s and excess reactant s the limiting reactant is o<sub>2</sub> since it would yield the least amount of product 18 mol br<sub>2</sub> the excess reactant is c<sub>2</sub>h<sub>3</sub>br<sub>3</sub> since its complete reaction would have yielded up to 54 mol br<sub>2</sub> think about your result

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web step 4 the reactant that produces a smaller amount of product is the limiting reagent mg mg produces less mgo mgo than does o<sub>2</sub> o<sub>2</sub> 3 98 g mgo vs 25 2 g mgo therefore mg is the limiting reagent in this reaction step 5 the reactant that produces a larger amount of product is the excess reagent

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web please save your changes before editing any questions 5 minutes 1 pt viewing model 3 if 4 moles of hydrogen react with 2 moles of oxygen which of these is the excess reactant is leftover at the end and how many moles of it are leftover hydrogen 2 moles leftover hydrogen 1 mole leftover

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web if you start with 14 82 g of  $\text{Ca(OH)}_2$  and 16 35 g of  $\text{H}_2\text{SO}_4$  a determine the limiting reagent b determine the number of moles of  $\text{H}_2\text{O}$  produced c determine the number of grams of  $\text{CaSO}_4$  produced d determine the number of grams of excess reagent left 1 make sure the equation is balanced

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