

Chapter 06 – Supplemental Practice Problems

1. An experiment calls for 24.0 moles of NH_3 . How many grams of NH_3 is this?

409 grams NH_3

2. A student was asked to prepare 6.84 grams of aspirin ($\text{C}_9\text{H}_8\text{O}_4$). How many moles of aspirin is this?

0.0380 moles aspirin

3. Calculate the number of grams in 0.125 moles of each of the following substances.

- a. Ammonium monohydrogen phosphate

16.5 grams $(\text{NH}_4)_2\text{HPO}_4$

- b. Calcium acetate

19.8 grams $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$

- c. Barium nitrate

32.7 grams $\text{Ba}(\text{NO}_3)_2$

- d. Hexanoic acid ($\text{C}_6\text{H}_{12}\text{O}_2$)

14.5 grams $\text{C}_6\text{H}_{12}\text{O}_2$

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Rosina Ehmman



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Pkg Acp Cer Gen Chem Ii Lab Chem 131 Book Review: Unveiling the Magic of Language

In an electronic digital era where connections and knowledge reign supreme, the enchanting power of language has are more apparent than ever. Its power to stir emotions, provoke thought, and instigate transformation is truly remarkable. This extraordinary book, aptly titled "**Pkg Acp Cer Gen Chem Ii Lab Chem 131**," compiled by a very acclaimed author, immerses readers in a captivating exploration of the significance of language and its profound impact on our existence. Throughout this critique, we shall delve into the book is central themes, evaluate its unique writing style, and assess its overall influence on its readership.

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