



## Chem 110B Practice Final Exam

### Part A. Please answer all questions thoroughly.

1. What energy equivalent appears above the arrow in the following calculated reaction?  
 $\text{p-NO}_2\text{Cl} \rightarrow \text{p-NO}_2\text{Cl} + \text{H}_2\text{O}$

eq 1                      eq 2                      eq 3                      eq 4                      eq 5                      \_\_\_\_\_

2. What energy equivalent appears below the arrow?

eq 6                      eq 7                      eq 8                      eq 9                      eq 10                      \_\_\_\_\_

3. What energy equivalent is below the arrow in the following reaction?

eq 11 or 12                      eq 13 or 14                      eq 15 or 16                      eq 17 or 18                      \_\_\_\_\_

4. What is the following reaction the product of the following reaction?

eq 19 or 20                      eq 21 or 22                      eq 23 or 24                      \_\_\_\_\_

5. What is the product of the reaction of the following with  $\text{H}_2\text{O}$ , assuming the reaction is carried out in the presence of  $\text{H}^+$ ?

eq 25 or 26                      eq 27 or 28                      eq 29 or 30                      eq 31 or 32                      \_\_\_\_\_

6. Complete the following table by supplying the chemical formula of the compound.

Source	Product
	$\text{PbCl}_2$
Calcium phosphate	
	$\text{Fe}_2\text{SO}_4$
	Calcium

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- Chapter 3: Pkg Acp Texas Tech Chem 1108 H in Everyday Life
- Chapter 4: Pkg Acp Texas Tech Chem 1108 H in Specific Contexts
- Chapter 5: Conclusion

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