

CHEM 1411 – General Chemistry I
Practice Problems, Chapters 1–3

Chapter 1 – Chemistry: The Study of Change

- 1.** Element, compound, homogeneous mixture (solution), or heterogeneous mixture:
- | | | |
|---------------------------------------|-----------------------|---|
| a) orange juice
(freshly-squeezed) | b) brass | c) 0.9% saline (NaCl) solution |
| d) garden soil | e) room air | f) methane gas |
| g) sodium metal | h) N ₂ gas | i) Cu(NO ₃) ₂ crystals |
- 2.** Define (some of these terms are found in Chapters 2 and 3):
- | | |
|-------------------------|--------------------------------|
| a) element | b) compound |
| c) solution | d) physical change |
| e) chemical change | f) law of conservation of mass |
| g) intensive properties | h) extensive properties |
| i) precision | j) accuracy |
| k) isotopes | l) stoichiometry |
| m) the mole | n) Avogadro's number |
| o) limiting reactant | |
- 3.** Convert
- | |
|--|
| a) 5 feet 10 inches to meters |
| b) 55 miles per hour to cm per second |
| c) 5.0 quarts to liters |
| d) 1.00 ft ² to cm ² |
| e) 32.06 amu to g |
| f) 25 g of carbon to moles of carbon |
| g) 1.00 g of gold to number of atoms of gold |
| h) 25 °F to Kelvins |
- 4.** How many significant figures are in the following numbers?
- | | | |
|-------------|--|---------------|
| a) 3.1416 | b) 6.022 × 10 ²³ atoms/mole | c) 0.000104 g |
| d) 6.0 g/mL | e) 1.00794 amu | f) 5000 mi |

Problems For General Chemistry Qualita

Sir Norman Lockyer



Problems For General Chemistry Qualita:

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1. This book is structured into several chapters, namely:

- Chapter 1: Introduction to Problems For General Chemistry Qualita
- Chapter 2: Essential Elements of Problems For General Chemistry Qualita
- Chapter 3: Problems For General Chemistry Qualita in Everyday Life
- Chapter 4: Problems For General Chemistry Qualita in Specific Contexts
- Chapter 5: Conclusion

2. In chapter 1, the author will provide an overview of Problems For General Chemistry Qualita. The first chapter will explore what Problems For General Chemistry Qualita is, why Problems For General Chemistry Qualita is vital, and how to effectively learn about Problems For General Chemistry Qualita.
3. In chapter 2, the author will delve into the foundational concepts of Problems For General Chemistry Qualita. The second chapter will elucidate the essential principles that must be understood to grasp Problems For General Chemistry Qualita in its entirety.
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6. In chapter 5, the author will draw a conclusion about Problems For General Chemistry Qualita. This chapter will summarize the key points that have been discussed throughout the book.

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