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oxidation $\text{Co} \rightarrow \text{Co}^{2+} + 2\text{e}^{-}$ reduction $\text{Fe}^{2+} + 2\text{e}^{-} \rightarrow \text{Fe}$ $24 \text{ 3 Ag} + \text{Ni} \rightarrow 3 \text{ Ag} + \text{Ni}^{2+}$ oxidation reduction $25 \text{ Cu} + \text{Pb} \rightarrow \text{Cu}^{2+} + \text{Pb}$ oxidation reduction 26 O_2

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