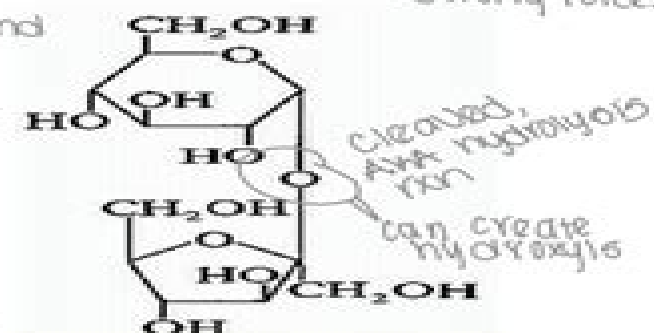


non covalent

- 5) Two macromolecules, such as proteins, can adhere tightly and specifically to each other. How can weak forces such as ionic bonds, van der Waals interactions, H-bonds, and hydrophobic forces lead to such strong adherence?
- A) adherence can be quite strong by having many weak forces involved *memoryize... multiple weak forces = 1 strong*
 - B) the weak forces can become very strong once non-polar groups are excluded from the inside of the molecules *adher.*
 - C) the weak forces are readily converted to covalent bonds, creating strong adherence between molecules *need enzyme to convert*
 - D) adhesion is strong because the weak forces can be involved in condensation reactions *only strong forces*

- 6) The structure at right is sucrose.
- A) Circle the bond which is cleaved by sucrase *glycosidic bond*
 - B) Is sucrose a reducing sugar? *NO. NO room for rxn.*



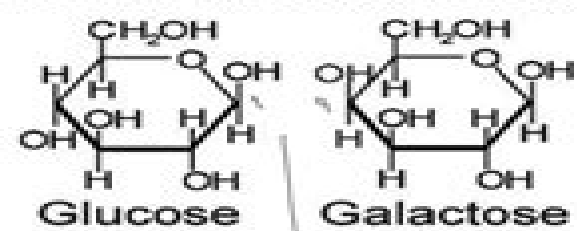
- 7) Routine examination of the urine of an asymptomatic pediatric patient showed a positive reaction with Clinistest (a copper reduction method of detecting reducing sugars), but a negative reaction to glucose oxidase test. Which one of the following sugars is least likely to be present (assuming a single elevated saccharide).
- A) Lactose
 - B) Fructose
 - C) Sucrose
 - D) Galactose

basically asking which is not reducing sugar.

- 8) Which of the following statements best describes glucose?
- A) It is a ketose and usually exists as a furanose ring in solution.
 - B) It is a C-4 epimer of galactose. *+ gluc - gluc?*
 - C) It is utilized in biological systems only in the L-isomeric form. *not true*
 - D) It is produced from dietary starch by the action of salivary α -amylase. *SUCROSE?*
 - E) Homopolysaccharides of glucose, formed by the action of glycosyltransferases, are always branched molecules that contain only β -glycosidic linkages. *1 trans or too?*

haven't covered yet so rule out

- 9) Lactose is a disaccharide of galactose linked to glucose via a β -1,4 linkage. Consulting the structures below, draw the structure of lactose. *not going to be asked on exam*



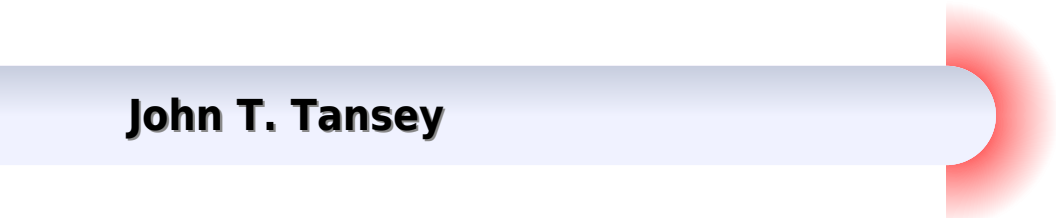
dehydration rxn? - H₂O



α L O J β

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