

HEAT Practice Problems

$$Q = m \times \Delta T \times C$$

1. 5.0 g of copper was heated from 20°C to 80°C. How much energy was used to heat Cu? (Specific heat capacity of Cu is 0.092 cal/g °C)
2. How much heat is absorbed by 20g granite boulder as energy from the sun causes its temperature to change from 10°C to 29°C? (Specific heat capacity of granite is 0.1 cal/g°C)
3. How much heat is released when 30 g of water at 96°C cools to 25°C? The specific heat of water is 1 cal/g°C.
4. If a 3.1g ring is heated using 10.0 calories, its temperature rises 17.9°C. Calculate the specific heat capacity of the ring.
5. The temperature of a sample of water increases from 20°C to 46.6°C as it absorbs 5650 calories of heat. What is the mass of the sample? (Specific heat of water is 1.0 cal/g °C)
6. The temperature of a sample of iron with a mass of 10.0 g changed from 50.4°C to 25.0°C with the release of 47 calories of heat. What is the specific heat of iron?
7. A 4.50 g coin of copper absorbed 54 calories of heat. What was the final temperature of the copper if the initial temperature was 25°C? The specific heat of copper is 0.092 cal/g°C.
8. A 155 g sample of an unknown substance was heated from 25°C to 40°C. In the process, the substance absorbed 569 calories of energy. What is the specific heat of the substance?
9. What is the specific heat of an unknown substance if a 2.50 g sample releases 12 calories as its temperature changes from 25°C to 20°C?

Problems In Heat And Thermodynamics

Josiah Willard Gibbs



Problems In Heat And Thermodynamics:

Worked Problems in Heat, Thermodynamics and Kinetic Theory for Physics Students L. Pincherle, 2013-10-22 Worked Problems in Heat Thermodynamics and Kinetic Theory for Physics Students is a complementary to textbooks in physics This book is a collection of exercise problems that have been part of tutorial classes in heat and thermodynamics at the University of London This collection of exercise problems with answers that are fully worked out deals with various topics This book poses problems covering the definition of temperature such as calculating the assigned value of the temperature of boiling water under specific conditions This text also gives example of problems dealing with the first law of thermodynamics and with the definition of thermal capacities Some practical questions such as problems dealing with thermal engines are presented This book then discusses problems using the energy equation as well as asking the student to derive a general equation of state of a material satisfying a specific condition This text challenges the student to use a T S diagram to calculate the efficiency of a reversible cycle under certain conditions Several other problems concern the Joule and Joule Kelvin effects low temperature physics and heat conduction This review material can be helpful for students of physics thermodynamics and related subjects It can also be used by teachers of physics

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Heat Transfer Kubie Jorge, Tariq Muneer, Grassie Thomas, 2012-08-06 A core task of engineers is to analyse energy related problems The analytical treatment is usually based on principles of thermodynamics fluid mechanics and heat transfer but is increasingly being handled computationally This unique resource presents a practical textbook written for both undergraduates and professionals with a series of over 60 computer workbooks on an accompanying CD The book emphasizes how complex problems can be deconstructed into a series of simple steps All thermophysical property computations are illustrated using diagrams within text and on the companion CD

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Solved Problems in Thermodynamics and Statistical Physics Gregor Skačej, Primož Zihnerl, 2019-11-09 This book contains a modern selection of about 200 solved problems and examples arranged in a didactic way for hands on experience with course work in a standard advanced undergraduate first year graduate class in thermodynamics and statistical physics The principles of thermodynamics and equilibrium statistical physics are few and simple but their application often proves more involved than it may seem at first sight This book is a comprehensive complement to any textbook in the field emphasizing the analogies between the different systems and paves the way for an in depth study of solid state physics soft matter physics and field theory

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Engineering Viktor Józsa, Róbert Kovács, 2019-10-24 This book provides general guidelines for solving thermal problems in the fields of engineering and natural sciences Written for a wide audience from beginner to senior engineers and physicists it provides a comprehensive framework covering theory and practice and including numerous fundamental and real world examples Based on the thermodynamics of various material laws it focuses on the mathematical structure of the continuum models and their experimental validation In addition to several examples in renewable energy it also presents thermal processes in space and summarizes size dependent non Fourier and non Fickian problems which have increasing practical relevance in e g the semiconductor industry Lastly the book discusses the key aspects of numerical methods particularly highlighting the role of boundary conditions in the modeling process The book provides readers with a comprehensive toolbox addressing a wide variety of topics in thermal modeling from constructing material laws to designing advanced power plants and engineering systems Cryogénie Pure Et Appliquée International Institute of Refrigeration, 1959

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Problems and Solutions in Waves, Heat and Thermodynamics, 2025-02-28 *Thermodynamics of Reversible Cycles in Gases and Saturated Vapors* Michael Pupin, 1894 *An Introduction to Thermal Physics* C. J. Adkins, 1987-02-05 This textbook is intended for introductory courses in physics engineering and chemistry at universities polytechnics and technical colleges It provides either an elementary treatment of thermal physics complete in itself for those who need to carry the subject no further or a sound foundation for further study in more specialised courses The author gives a clear and concise account of those basic concepts that provide the foundations for an understanding of the thermal properties of matter The area covered corresponds very roughly to the traditional topics of heat kinetic theory and those properties of matter for which there are elementary explanations in terms of interatomic forces The book is not concerned with experimental detail but with ideas and concepts and their quantitative application through simple models The author provides many problems for which the answers are included The book should also be useful in teacher training and as a reference book in the libraries of schools where pupils are being prepared for tertiary courses

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