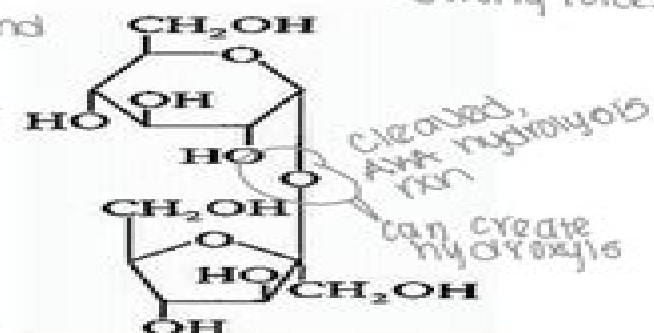


# non covalent

- 5) Two macromolecules, such as proteins, can adhere tightly and specifically to each other. How can weak forces such as ionic bonds, van der Waals interactions, H-bonds, and hydrophobic forces lead to such strong adherence?
- A) adherence can be quite strong by having many weak forces involved *memoryize... multiple weak forces = 1 strong*
  - B) the weak forces can become very strong once non-polar groups are excluded from the inside of the molecules *hydrophobic*
  - C) the weak forces are readily converted to covalent bonds, creating strong adherence between molecules *need enzyme to convert*
  - D) adhesion is strong because the weak forces can be involved in condensation reactions *only strong forces*

- 6) The structure at right is sucrose.
- A) Circle the bond which is cleaved by sucrase *glycosidic bond*
  - B) Is sucrose a reducing sugar? *NO. NO room for rxn.*



basically asking which is not reducing sugar.

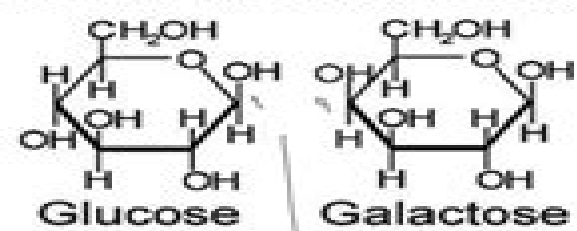
- 7) Routine examination of the urine of an asymptomatic pediatric patient showed a positive reaction with Clinistest (a copper reduction method of detecting reducing sugars), but a negative reaction to glucose oxidase test. Which one of the following sugars is least likely to be present (assuming a single elevated saccharide).
- A) Lactose
  - B) Fructose
  - C) Sucrose
  - D) Galactose

haven't covered yet so rule out

- 8) Which of the following statements best describes glucose?
- A) It is a ketose and usually exists as a furanose ring in solution. *also sugar*
  - B) It is a C-4 epimer of galactose. *gluc - gluc?*
  - C) It is utilized in biological systems only in the L-isomeric form. *not true*
  - D) It is produced from dietary starch by the action of salivary  $\alpha$ -amylase. *SUCROSE?*
  - E) Homopolysaccharides of glucose, formed by the action of glycosyltransferases, are always branched molecules that contain only  $\beta$ -glycosidic linkages. *1 trans or too?*

lippincott's question.

- 9) Lactose is a disaccharide of galactose linked to glucose via a  $\beta$ -1,4 linkage. Consulting the structures below, draw the structure of lactose. *not going to be asked on exam*



dehydration rxn? - H<sub>2</sub>O



$\alpha$  L O J  $\beta$

# Problems In Biochemistry

**E. A. Dawes**



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*Biochemistry* John T. Tansey, 2019-04-02 Biochemistry An Integrative Approach is addressed to premed biochemistry and life science majors taking a one semester biochemistry course This version includes the first 12 chapters and should only be used for one semester biochemistry courses Biochemistry addresses the diverse needs of premed biochemistry and life science majors by presenting relevant material while still preserving a chemical perspective Presented within the next generation of WileyPLUS Biochemistry emphasizes worked problems through video walkthroughs interactive elements and expanded end of chapter problems with a wide range of subject matter and difficulty The worked problems in the course are both qualitative and quantitative and model for students the biochemical reasoning they need to practice Students will often be asked to analyze data and make critical assessments of experiments

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**Fundamentals of Biochemical Calculations, Second Edition** Krish Moorthy,2008 Fundamentals of Biochemical Calculations Second Edition demystifies the fundamental calculations used in modern biochemistry cell biology and allied biomedical sciences The book encouragesbothundergraduates and scientists to develop an understanding of the processes involved in performing biochemical calculations rather than rely on memorized formulae Detailed calculation steps useful comments and mathematical hints enhance solved problems throughout Encouraging the use of the Ratio Method the author draws attention to its applicability to a wide variety of biochemistry problems with logical consistency and intuitive ease This entirely revised and updated second edition introduces three new chapters with coverage of laboratory oriented calculations and data presentation calculations relating to the emerging techniques in DNA and molecular biology and applying the Ratio Method to pharmaceutical calculations New edition adds questions and extra tips in every chapter to help students perform biochemical calculations on their own By strengthening their understanding of mathematical logic and problem solving skills Fundamentals of Biochemical Calculations Second Edition enables students and biochemists to explore a wider range of real world applications with the ability to tackle increasingly complex problems in biochemistry and related fields     The American Naturalist ,1897     **Kretovich, V. L.: Biochemical principles of the food industry** ,1963

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