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**Mark Howard**



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*The Social-Structural Determinants of Health* Sofie Hass,Jessica R. Corno,Aida Wisecup,2025-07-29 Today more than ever society is faced with the unequivocal question of developing a better understanding of the interconnectedness between social forces environmental experiences and the outcomes of mental and physical health and well being Post COVID 19 considerations have led to an unprecedented need to establish a more interdisciplinary understanding of the role that social construction has on health and illness and the diverse set of ways that our social world plays in the identification treatment and

experience of various diseases and health related conditions Social life is intricately linked to quality of life variables related to health both mental and physical Thus we recognize the invaluable contribution to the field of medicine made by a comprehensive understanding of health wellness and social functioning as a byproduct of the study of the human body and mind However we posit that this knowledge would be incomplete without examining the close relationship between health wellness and those aspects of social environment that create conditions that can either strengthen or undermine the wellbeing of individuals groups of people and entire communities the social structural determinants of health

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web we then changed the volume between 5ml and 20ml and recorded the pressure at each this part of the lab showed that volume and pressure are inversely proportionate as the pressure would decrease if the volume was increased part b of the experiment looked at the relationship between temperature and pressure

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web procedure 1 the relationship between pressure and volume set the plunger of the syringe to 1/10 the capacity of the syringe the 20 ml mark if using a 20 ml syringe the volume is read from the edge of the bottom rib of the plunger attach the syringe to the connector at the end of sensor

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web he concluded that at a constant pressure the volume of the gas would expand at a particular rate for each degree of temperature rise that being 1/267 per degree in 1874 Victor Regnault obtained better experimental results showing this number to be 1/273 and concluded that the pressure would approach zero at 1/273 degrees C

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web the volume of 1.00 mol of any gas at STP standard temperature 273.15 K and pressure 1 atm is measured to be

22 414 l 22 414 l we can substitute 101 325 kpa 101 325 kpa for pressure 22 414 l 22 414 l for volume and 273 15 k 273 15 k for temperature into the ideal gas equation and solve for r r

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web may 20 2018 french physicist jacques charles 1746 1823 studied the effect of temperature on the volume of a gas at constant pressure charles s law states that the volume of a given mass of gas varies directly with the absolute temperature of the gas when pressure is kept constant the absolute temperature is temperature measured

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web chemists sometimes make comparisons against a standard temperature and pressure stp for reporting properties of gases 273 15 k and 1 atm 101 325 kpa 1 at stp one mole of an ideal gas has a volume of about 22 4 l this is referred to as the standard molar volume figure 9 18

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web this lab explores the relationships between pressure volume and temperature in an ideal gas the ideal gas law  $pV = nRT$  is used and with it we can explore it by seeing how the pressure and volumes changed in the tubes each time that we

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web figure page index 10 since the number of moles in a given volume of gas varies with pressure and temperature changes chemists use standard temperature and pressure 273 15 k and 1 atm or 101 325 kpa to report properties of gases

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web pressure increases even though there were errors in the experiment the goal of the experiment was met the goal of declaring what type of relationship it is was clear using the data pressure kpa temperature c temperature k constant k p t 95 82 kpa 16 0 c 289 k 0 33 97 90 kpa 19 0 c 292 k 0 34 98 97 kpa 99 22 kpa 101 4 kpa 19 1 c

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web the properties of water at constant volume can be represented as a function of pressure and temperature as shown in the diagram below figure 1 the pressure temperature relationship at constant volume the saturation point of water is the condition at which a phase change occurs from liquid to vapour or vapour to liquid

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