

**Claim:** Unknown solution #5 is potassium chromate, or  $K_2CrO_4$ .

**Evidence:** Unknown solution #5 turned purple in the flame test, just like potassium turned purple. It also changed a darker color and got moderately cloudy from the reactions of  $HCl$  and  $BaCl_2$ , just like chromate did.

**Reasoning:** Because the reactions from the  $HCl$  and  $BaCl_2$  and the flame test was the same as potassium and chromate, unknown solution #5 has to be  $K_2CrO_4$ , or Potassium Chromate.

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