

CHEM 101 Formula & Data Sheet

1A (1)																	2A (2)											3A (13)	4A (14)	5A (15)	6A (16)	7A (17)	8A (18)		
1 H 1.008																	2 He 4.003											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 19.00	10 Ne 20.18		
2 Li 6.941	3 Be 9.012																	11 Na 22.99	12 Mg 24.31											13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.06	17 Cl 35.45	18 Ar 39.95
TRANSITION ELEMENTS																																			
3 Na 22.99	4 Mg 24.31	5 Sc (45)	6 Ti (48)	7 V (51)	8 Cr (52)	9 Mn (55)	10 Fe (56)	11 Co (59)	12 Ni (59)	13 Cu (63)	14 Zn (65)	15 Ga 69.72	16 Ge 72.63	17 As 74.92	18 Se 78.96	19 Br 79.90	20 Kr 83.80																		
4 K 39.09	5 Ca 40.08	6 Sc 44.96	7 Ti 47.87	8 V 50.94	9 Cr 52.00	10 Mn 54.94	11 Fe 55.85	12 Co 58.93	13 Ni 58.69	14 Cu 63.55	15 Zn 65.38	16 Ga 69.72	17 Ge 72.63	18 As 74.92	19 Se 78.96	20 Br 79.90	21 Kr 83.80																		
5 Rb 85.47	6 Sr 87.62	7 Y 88.91	8 Zr 91.22	9 Nb 92.91	10 Mo 95.96	11 Tc (98)	12 Ru 101.1	13 Rh 102.9	14 Pd 106.4	15 Ag 107.9	16 Cd 112.4	17 In 114.8	18 Sn 118.7	19 Sb 121.8	20 Te 127.6	21 I 126.9	22 Xe 131.3																		
6 Cs 132.9	7 Ba 137.3	8 La 138.9	9 Hf 178.5	10 Ta 180.9	11 W 183.8	12 Re 186.2	13 Os 190.2	14 Ir 192.2	15 Pt 195.1	16 Au 197.0	17 Hg 200.6	18 Tl 204.4	19 Pb 207.2	20 Bi 209.0	21 Po (209)	22 At (210)	23 Rn (222)																		
7 Fr (223)	8 Ra (226)	9 Ac (227)	10 Th (232)	11 Pa (231)	12 U (238)	13 Np (237)	14 Pu (244)	15 Am (243)	16 Cm (247)	17 Bk (247)	18 Cf (251)	19 Es (252)	20 Fm (257)	21 Md (258)	22 No (259)	23 Lr (262)	24 Og (264)																		

6	Lanthanides	58 Ce 140.1	59 Pr 140.9	60 Nd 144.2	61 Pm (145)	62 Sm 150.4	63 Eu 152.0	64 Gd 157.3	65 Tb 158.9	66 Dy 162.5	67 Ho 164.9	68 Er 167.3	69 Tm 168.9	70 Yb 173.1	71 Lu 175.0
7	Actinides	90 Th 232.0	91 Pa (231)	92 U 238.0	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

Formulas

$$E = h\nu = \frac{hc}{\lambda} \quad c = \lambda \nu \quad E_n = -R_H \left(\frac{Z^2}{n^2} \right) \quad \Delta E_n = -R_H Z^2 \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$$

$$FC = NVE - (\text{unshared } e^- + \frac{1}{2} \text{ shared } e^-) \quad BO = \frac{1}{2} (\text{bonding } e^- - \text{antibonding } e^-)$$

$$\Delta H^\circ_{rxn} \approx \sum BE(\text{bonds broken}) - \sum BE(\text{bonds formed}) \quad \text{Electrostatic force} = -k \left(\frac{q_1 q_2}{r} \right)$$

$$(KE)_{avg} = \frac{3}{2} RT \quad u_{rms} = \sqrt{\frac{3RT}{M}} \quad P_i = x_i P_T \quad x_i = \frac{n_i}{n_{tot}} = \frac{P_i}{P_{tot}}$$

$$PV = nRT \quad (P + \frac{an^2}{V^2})(V - nb) = nRT$$

$$V(\text{sphere}) = \frac{4}{3} \pi r^3 \quad V(\text{cube}) = \ell^3 \quad \text{BCC: } \ell = \frac{4r}{\sqrt{3}} \quad \text{FCC: } \ell = \sqrt{8} r$$

Constants and Conversion Factors

$N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$	$1 \text{ atm} = 760 \text{ mmHg} = 760 \text{ torr} = 101.3 \text{ kPa}$
$h = 6.626 \times 10^{-34} \text{ J s}$	$0.0^\circ \text{C} = 273.15 \text{ K}$
$c = 2.998 \times 10^8 \text{ m s}^{-1}$	$1 \text{ cal} = 4.184 \text{ J}$
$R_H = 2.179 \times 10^{-18} \text{ J}$	$1 \text{ eV} = 1.602 \times 10^{-19} \text{ J} = 96.5 \text{ kJ mol}^{-1}$
$R = 0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1}$	$1 \text{ nm} = 10^{-9} \text{ m}$
$= 0.08314 \text{ L bar mol}^{-1} \text{ K}^{-1}$	$1 \text{ \AA} = 10^{-10} \text{ m}$
$= 8.314 \text{ J mol}^{-1} \text{ K}^{-1}$	$1 \text{ pm} = 10^{-12} \text{ m}$

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