

CHEM 1407: Isomers and Nomenclature Lab Activity

What are isomers?

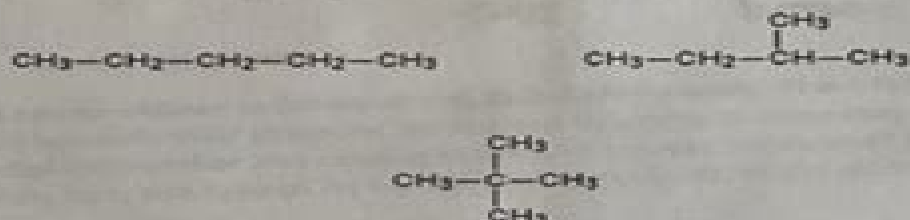
Isomers are molecules that have the same molecular formula, but have a different arrangement of the atoms in space.

You will study: Constitutional Isomers and Stereoisomers

In structural /constitutional isomerism, the atoms are arranged in a completely different order.

How are these constitutional isomers different from each other?

A. The number of carbons in the longest chain continuous chain is different.



B. The longest chain remains the same but the position of functional groups, side groups or multiple bonds change.



C. Isomers contain different functional groups



How are stereoisomers different? Two major types- diastereomers and enantiomers.

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web in each of the following redox reactions indicate the elements that undergo changes in oxidation number and the magnitude of change then identify the element oxidized and reduced as well as the oxidizing agent and reducing agent

$$\text{a) } \text{I}_2\text{O}_5 + 5\text{CO} \rightarrow \text{I}_2 + 5\text{CO}_2$$

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web chemical reactions in which there is a transfer of electrons from one substance to another are known as oxidation reduction reactions or redox reactions in this unit you will examine the oxidation reduction process and use the oxidation state and oxidation number concepts to identify redox reactions and to keep track of electrons

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web jun 28 2010 answers i determine what is oxidized and what in each reaction identify the oxidizing agent and the reducing agent also

$$2\text{Sr} + \text{O}_2 \rightarrow 2\text{SrO}$$

Sr 0 to Sr 2 oxidized reducing agent O 0 to O 2 reduced ox ag 2Li + S → Li₂S

Li 0 to Li 1 oxidized red ag S 0 to S 2 reduced ox ag 3 2Cs + Br₂