

# CHEM 1405 Practice Exam # 4 (Ch 8-10)

Name \_\_\_\_\_

Section # \_\_\_\_\_

**MULTIPLE CHOICE.** Choose the one alternative that best completes the statement or answers the question.

- 1) Which one of the following is characteristic of a base? 1) \_\_\_\_\_
  - A) has a sour taste
  - B) is insoluble in water
  - C) produces  $\text{H}_3\text{O}^+$  in water
  - D) turns blue litmus red
  - E) has a slippery, soapy feel
  
- 2) Which of the following is correctly identified? 2) \_\_\_\_\_
  - A)  $\text{HCl}$ , weak acid
  - B)  $\text{Ca}(\text{OH})_2$ , weak base
  - C)  $\text{NaOH}$ , strong base
  - D)  $\text{NH}_3$ , strong acid
  - E)  $\text{H}_2\text{CO}_3$ , strong acid
  
- 3) The conjugate base of  $\text{H}_2\text{O}$  is \_\_\_\_\_. 3) \_\_\_\_\_
  - A)  $\text{OH}^-$
  - B)  $\text{O}^{2-}$
  - C)  $\text{H}_3\text{O}^+$
  - D)  $\text{H}_2\text{O}$
  - E)  $\text{H}_2\text{O}$  has no conjugate base.
  
- 4) Which of the following is the strongest acid? 4) \_\_\_\_\_
  - A)  $\text{H}_3\text{PO}_4$
  - B)  $\text{NH}_4^+$
  - C)  $\text{H}_2\text{CO}_3$
  - D)  $\text{HCl}$
  - E)  $\text{NaOH}$
  
- 5) When a reaction is at equilibrium, \_\_\_\_\_. 5) \_\_\_\_\_
  - A) the reaction is no longer reversible
  - B) the products and reactants have the same energy content
  - C) no more reactants are converted to products
  - D) the forward and reverse reactions occur at the same rate
  - E) all reaction stops
  
- 6) What is the pH of a solution with  $[\text{OH}^-] = 1 \times 10^{-4} \text{ M}$ ? 6) \_\_\_\_\_
  - A) -10.0
  - B) 10.0
  - C) 4.0
  - D) -4.0
  - E)  $1.0 \times 10^{-10}$

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