

## ACTIVITY 5

COMPOUND IDENTIFICATION OF UNKNOWN  
DRUG SAMPLES USING PHYSICAL PROPERTIES

## PHYSICAL PROPERTIES

## 1. QUALITY ASSURANCE

- Detection of physical changes
  - Presence of all contaminants and instability are a sign of poor quality

## 2. STABILITY ASSESSMENT

- Monitor and assess stability
  - Emulsion (separation of oil and water) is a detection of instability

## 3. EARLY DETECTION OF CONTAMINATION

- Reveal the presence of contaminants/impurities
  - The formation or presence of molds and/or precipitations is an early detection of contamination

## 4. BATCH TO BATCH CONSISTENCY

- Same quality within batches
  - Batch no. - series of numbers
  - Lot no. - portion of the batch

## SOLUBILITY

## PRINCIPLE

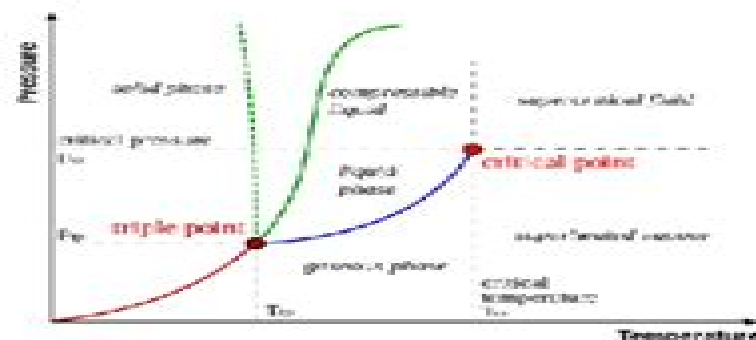
- Like dissolves like meaning polar solutes dissolve in polar solvents
- Solubility product ( $K_{sp}$ )
  - Solubility product constant
  - indication to determine if a sample is contaminated but it should not be the sole basis to determine it
  - $K_{sp}$  values are found in pharmacopoeias
  - $S_x = 2.00 \rightarrow K_{sp} = 1.6$ 
    - It means that  $S_x$  is contaminated because it does not match the one in the pharmacopoeia
- What is reflected in the USP are pure substances

## PURPOSE

- Formulation Development
  - Properly formulate drugs and maximize intended effects by incorporating excipients
- Drug Delivery Systems
  - Desired drug release profiles
- Bioavailability
  - Solubility and absorption

## MELTING POINT RANGE

"The temperature from the point at which solid first begins to liquify to the point at which the entire sample is liquid."



## PRINCIPLE

- Eutectic Point (Solid to Liquid)
  - Lowest temperature where phase transformation is observed
- Phase Transition (Solid to Liquid)
  - Very narrow temperature range

## Kinetic Energy and IMFA

- Inversely proportional (increased KE, decreased IMFA)

## Triple Point

- Where solid, liquid and gas will exist simultaneously

## Critical Point

- Difficult for liquid to be converted into gas state

## Supercritical Fluid

- Impossible to convert liquid into its gas state (high pressure)

# Pkgacp Chemistry 113l Lab Manual

**Sapp**



## **Pkgacp Chemistry 113l Lab Manual:**

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*Laboratory Manual for Principles of General Chemistry* Jo Allan Beran, 1999-11-17 This flexible lab manual appropriate for use with a wide range of general chemistry books offers a wealth of practical chemistry experiments It includes pertinent information on rules and safety in the lab Preparation of the new edition was guided by specific feedback from users

Laboratory Manual for Principles of General Chemistry Jo Allan Beran, 2004-04-15 This remarkably popular lab manual has won over users time and time again with its exceedingly clear presentation and broad selection of topics and experiments Now revised and fine tuned this new Seventh Edition features three new experiments Water Analysis Solids Experiment 3 Vitamin C Analysis Experiment 16 and Hard Water Analysis Experiment 30 In addition nearly 90% of the Prelaboratory Assignment Questions and Laboratory Questions are either new or revised

*Chemistry* Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman, 2004-06-30

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*Laboratory Manual of General Chemistry* Arthur Becket Lamb, 1916

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**Laboratory Manual for Principles of General Chemistry** Foster, 2004-01-01

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