

2020-2021 Chemical Engineering
Examination 1, August 20, 2020

Name: _____

Name: _____

Thermodynamic properties of water vapor at 10 bar (sat) are given in the table below in SI units, where h is in kJ kg^{-1} , u is in kJ kg^{-1} , s is in $\text{kJ kg}^{-1} \text{K}^{-1}$, and c_p is in $\frac{\text{kJ}}{\text{kg K}}$.

The following are assumed to be true:

1. Multiple choice questions 1-4 have one correct answer. Mark the answer provided by the production designer, provided that both other questions 2) or another answer is provided. 2) or the correct 3) and 4) are given in every question.

For questions 1-4, mark the following statements. If the statement is true, mark "True" and if the statement is false, mark "False". If the statement is false, mark "False" and if the statement is true, mark "True".

Water: $c_p = 4.18 \text{ kJ kg}^{-1} \text{K}^{-1}$, $h_{\text{sat}} = 2675.5 \text{ kJ kg}^{-1}$, $u_{\text{sat}} = 2029.6 \text{ kJ kg}^{-1}$, $s_{\text{sat}} = 7.3549 \text{ kJ kg}^{-1} \text{K}^{-1}$.

1) The specific enthalpy of water vapor at 10 bar (sat) is $2675.5 \text{ kJ kg}^{-1}$. Is this statement true or false?

- a. True ☐ b. False ☐ c. True ☐ d. False ☐
 e. True ☐ f. False ☐ g. True ☐ h. False ☐

2) The mass of water vapor produced in the water vaporization process is:

- a. 10 kg ☐ b. 10 kg ☐ c. 10 kg ☐ d. 10 kg ☐
 e. 10 kg ☐ f. 10 kg ☐ g. 10 kg ☐ h. 10 kg ☐

3) The specific enthalpy of water vapor at 10 bar (sat) is $2675.5 \text{ kJ kg}^{-1}$. Is this statement true or false?

Statement	True/False	Answer (True/False)
1. True	True/False	True/False
2. True	True/False	True/False
3. True	True/False	True/False

4) The specific enthalpy of water vapor at 10 bar (sat) is $2675.5 \text{ kJ kg}^{-1}$. Is this statement true or false?

- a. True ☐ b. False ☐ c. True ☐ d. False ☐
 e. True ☐ f. False ☐ g. True ☐ h. False ☐

5) The specific enthalpy of water vapor at 10 bar (sat) is $2675.5 \text{ kJ kg}^{-1}$. Is this statement true or false?

- a. True ☐ b. False ☐ c. True ☐ d. False ☐

Pkgacp Chem 143

Sasha Hu



Pkgacp Chem 143:

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