

# CHEM 110 SPECTROSCOPY DATA SHEET

IMPORTANT INFRARED ABSORPTION REGIONS (STRETCHING FREQUENCIES)		
Bond Present		$\bar{\nu}$ (cm <sup>-1</sup> )
O-H	(alcohol, phenol)	3700 – 3200
	(carboxylic acid)	3600 – 2500 (broad)
N-H		3500 – 3100
C-H		3100 – 2800
C=O		1850 – 1600
C=C	(aliphatic)	1680 – 1600
	(aromatic)	1620 – 1550
C-O		1250 – 1050 (less definitive)

<sup>1</sup> H NUCLEAR MAGNETIC RESONANCE SPECTROSCOPY ( <sup>1</sup> H NMR) Approximate Chemical Shifts ( $\delta$ ) of Hydrogens in Organic Compounds		
	Hydrogens ( <i>H</i> ) in various chemical environments	$\delta$ (ppm downfield from TMS)
Hydrogens bonded to <i>sp</i> <sup>3</sup> carbon	(CH <sub>3</sub> ) <sub>4</sub> Si (TMS)	0
	R <sub>2</sub> CH $\overset{ }{\underset{ }{\text{C}}}$ R	1.0 (approx.)
	R <sub>2</sub> CH $\overset{ }{\underset{ }{\text{C}}}$ Z*	1.0 - 2.0
	R <sub>2</sub> CH $\overset{ }{\underset{ }{\text{C}}}=\overset{ }{\text{C}}-$	1.6 - 1.9
	R <sub>2</sub> CH $\overset{ }{\underset{ }{\text{C}}}=\text{O}$	2.0 - 2.5
	R <sub>2</sub> CHAr	2.5 - 3.5
	R <sub>2</sub> CHX**	2.0 - 3.7
	R <sub>2</sub> CHO-	3.3 - 4.5
Hydrogens bonded to <i>sp</i> <sup>2</sup> carbon	H $\overset{ }{\underset{ }{\text{C}}}=\overset{ }{\text{C}}-$	4.5 - 7.0
	H-Ar	6.5 - 8.0
	H $\overset{ }{\underset{\text{O}}{\text{C}}}$ R	9.5 - 10.0
Hydrogen bonded to oxygen	HOR	3.0 - 6.0 (variable)
	H $\overset{ }{\underset{\text{O}}{\text{C}}}$ R	10.5 - 12.0

Notes:

\*\*X = Cl, Br, I or the N of an amine    \*Z = O, C=O, or X (as defined above)

R = H or alkyl group

# Pkg Acp Chem 110

**CH Cherryholmes**



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