

Chemistry 235 Lab Final Pre and Post Lab Questions and answers | updated 2024

Exp. 1 define the melting point of a substance. - it is the temperature at which a solid becomes a liquid.

Exp. 1 why should you always use a new capillary tube when performing a second melting point determination on your sample? - to keep the sample from getting impurities

Exp. 1 why is it necessary to allow the Mel-Temp to cool before performing a second melting point determination? - a rapid temperature increase on the substance being tested will give a melting point range that is too high.

Exp. 1 in the experiment you used only 2-3 mm of sample in the melting point tube.

what disadvantage is associated with using too much material? - if there is too much material, the melting point range is affected and changed. it would be broader than expected

Exp. 1 should you use the same melting point apparatus for all melting point measurements or is it okay to use a different one for each measurement? explain. - in the laboratory, all the Mel-Temp apparatuses were the same, so as long as they were cool enough to use, it is alright to use different ones.

Exp. 1 the Mel-Temp method is generally not useful for determining the melting point of inorganic compounds. explain why. - inorganic compounds have extremely high melting points. they have strong intermolecular forces, causing them to have very high melting points. the Mel-Temp equipment would not be able to supply enough temperature to force the inorganic compounds to melt.

Exp. 2 dimethylsulfoxide (DMSO, bp 189 degrees celsius) is an excellent organic solvent, but is not routinely employed in crystallization. explain why this might be - in can dissolve substances extremely well. however, it can also be potentially dangerous to work with as it is slightly toxic

Exp. 2 if you perform a recrystallization on 435 mg of impure benzoic acid and recover 360 mg of crystals, what is the percent recovery? - $360/435 \times 100 = 82.75\%$

Exp. 2 what will happen if the centrifuge is not properly balanced? - if a centrifuge is not properly balanced, it can injure you and ruin the centrifuge as well. they operate at high speeds and unbalanced can result in the sample container breakage, which can be potentially harmful depending on the sample type.

Exp. 2 list some advantages as well as disadvantages to using the Craig tube crystallization technique - the Craig tube method is good for relatively small amounts of material (10-100mg). it also provides a better percent recovery. however, it does not work with large amounts of material.

Exp. 2 what was the purpose of using activated charcoal in this experiment? did you notice any differences before and after the addition of the activated charcoal? - activated charcoal works to remove colored impurities by allowing the colored molecules to stick to the surface of the activated charcoal, which is also a form of

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Table of Contents **Pkg Acp Cer Chemistry 235**

1. Understanding the eBook **Pkg Acp Cer Chemistry 235**
 - The Rise of Digital Reading **Pkg Acp Cer Chemistry 235**
 - Advantages of eBooks Over Traditional Books
2. Identifying **Pkg Acp Cer Chemistry 235**
 - Exploring Different Genres
 - Considering Fiction vs. Non-Fiction
 - Determining Your Reading Goals
3. Choosing the Right eBook Platform
 - Popular eBook Platforms
 - Features to Look for in an **Pkg Acp Cer Chemistry 235**
 - User-Friendly Interface
4. Exploring eBook Recommendations from **Pkg Acp Cer Chemistry 235**
 - Personalized Recommendations
 - **Pkg Acp Cer Chemistry 235** User Reviews and Ratings
 - **Pkg Acp Cer Chemistry 235** and Bestseller Lists

-
5. Accessing Pkg Acp Cer Chemistry 235 Free and Paid eBooks
 - Pkg Acp Cer Chemistry 235 Public Domain eBooks
 - Pkg Acp Cer Chemistry 235 eBook Subscription Services
 - Pkg Acp Cer Chemistry 235 Budget-Friendly Options
 6. Navigating Pkg Acp Cer Chemistry 235 eBook Formats
 - ePub, PDF, MOBI, and More
 - Pkg Acp Cer Chemistry 235 Compatibility with Devices
 - Pkg Acp Cer Chemistry 235 Enhanced eBook Features
 7. Enhancing Your Reading Experience
 - Adjustable Fonts and Text Sizes of Pkg Acp Cer Chemistry 235
 - Highlighting and Note-Taking Pkg Acp Cer Chemistry 235
 - Interactive Elements Pkg Acp Cer Chemistry 235
 8. Staying Engaged with Pkg Acp Cer Chemistry 235
 - Joining Online Reading Communities
 - Participating in Virtual Book Clubs
 - Following Authors and Publishers Pkg Acp Cer Chemistry 235
 9. Balancing eBooks and Physical Books Pkg Acp Cer Chemistry 235
 - Benefits of a Digital Library
 - Creating a Diverse Reading Collection Pkg Acp Cer Chemistry 235
 10. Overcoming Reading Challenges
 - Dealing with Digital Eye Strain
 - Minimizing Distractions
 - Managing Screen Time
 11. Cultivating a Reading Routine Pkg Acp Cer Chemistry 235
 - Setting Reading Goals Pkg Acp Cer Chemistry 235
 - Carving Out Dedicated Reading Time
 12. Sourcing Reliable Information of Pkg Acp Cer Chemistry 235
 - Fact-Checking eBook Content of Pkg Acp Cer Chemistry 235
 - Distinguishing Credible Sources
 13. Promoting Lifelong Learning

- Utilizing eBooks for Skill Development
 - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
 - Interactive and Gamified eBooks

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