

22FICMP Principles of Chemistry I (CHE-121-20101)

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Supporting for:	Thermodynamics, Equilibrium, Kinetics, Electrochemistry
Name:	Thermodynamics
Concepts covered for:	Thermodynamics, Equilibrium, Kinetics, Electrochemistry
Other subjects:	Chemistry, Physics
Course:	Chemistry 121-20101

Question 1

Answer:

Mark: 1.00 out of 1.00
Feedback:

Question 1

The following information was obtained from a graph of $\ln K$ vs. $1/T$ (K⁻¹). Which equation best describes the relationship between $\ln K$ and $1/T$?

Answer:

- ☐ $\ln K = -\frac{E_a}{R} \left(\frac{1}{T} - \frac{1}{T_0} \right)$
- ☐ $\ln K = -\frac{E_a}{R} \left(\frac{1}{T} \right) + \ln A$
- ☐ $\ln K = -\frac{E_a}{R} \left(\frac{1}{T} \right) + \ln A$
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- ☐ $\ln K = -\frac{E_a}{R} \left(\frac{1}{T} \right) + \ln A$

Feedback:

The correct answer is: $\ln K = -\frac{E_a}{R} \left(\frac{1}{T} \right) + \ln A$

Question 2

Answer:

Mark: 1.00 out of 1.00

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