

Chapter 06 – Supplemental Practice Problems

1. An experiment calls for 24.0 moles of NH_3 . How many grams of NH_3 is this?

409 grams NH_3

2. A student was asked to prepare 6.84 grams of aspirin ($\text{C}_9\text{H}_8\text{O}_4$). How many moles of aspirin is this?

0.0380 moles aspirin

3. Calculate the number of grams in 0.125 moles of each of the following substances.

- a. Ammonium monohydrogen phosphate

16.5 grams $(\text{NH}_4)_2\text{HPO}_4$

- b. Calcium acetate

19.8 grams $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$

- c. Barium nitrate

32.7 grams $\text{Ba}(\text{NO}_3)_2$

- d. Hexanoic acid ($\text{C}_6\text{H}_{12}\text{O}_2$)

14.5 grams $\text{C}_6\text{H}_{12}\text{O}_2$

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