

1. Ethanol reacts with sodium metal to form sodium ethoxide and hydrogen gas. The balanced chemical equation is:

- a. $2\text{C}_2\text{H}_5\text{OH} + 2\text{Na} \rightarrow 2\text{C}_2\text{H}_5\text{ONa} + \text{H}_2$
- b. $\text{C}_2\text{H}_5\text{OH} + \text{Na} \rightarrow \text{C}_2\text{H}_5\text{ONa} + \text{H}_2$
- c. $\text{C}_2\text{H}_5\text{OH} + \text{Na} \rightarrow \text{C}_2\text{H}_5\text{ONa} + \text{H}_2$
- d. $\text{C}_2\text{H}_5\text{OH} + \text{Na} \rightarrow \text{C}_2\text{H}_5\text{ONa} + \text{H}_2$

2. Which of the following is the correct chemical formula for sodium ethoxide?

- a. $\text{C}_2\text{H}_5\text{ONa}$
- b. $\text{C}_2\text{H}_5\text{O}_2\text{Na}$
- c. $\text{C}_2\text{H}_5\text{ONa}$
- d. $\text{C}_2\text{H}_5\text{ONa}$

3. In the reaction of ethanol with sodium metal, the sodium metal is oxidized to sodium ethoxide. The balanced chemical equation is:

- a. $2\text{C}_2\text{H}_5\text{OH} + 2\text{Na} \rightarrow 2\text{C}_2\text{H}_5\text{ONa} + \text{H}_2$
- b. $\text{C}_2\text{H}_5\text{OH} + \text{Na} \rightarrow \text{C}_2\text{H}_5\text{ONa} + \text{H}_2$
- c. $\text{C}_2\text{H}_5\text{OH} + \text{Na} \rightarrow \text{C}_2\text{H}_5\text{ONa} + \text{H}_2$
- d. $\text{C}_2\text{H}_5\text{OH} + \text{Na} \rightarrow \text{C}_2\text{H}_5\text{ONa} + \text{H}_2$

4. Which of the following is the correct chemical formula for sodium ethoxide?

- a. $\text{C}_2\text{H}_5\text{ONa}$
- b. $\text{C}_2\text{H}_5\text{O}_2\text{Na}$
- c. $\text{C}_2\text{H}_5\text{ONa}$
- d. $\text{C}_2\text{H}_5\text{ONa}$

5. Which of the following is the correct chemical formula for sodium ethoxide?

- a. $\text{C}_2\text{H}_5\text{ONa}$
- b. $\text{C}_2\text{H}_5\text{O}_2\text{Na}$
- c. $\text{C}_2\text{H}_5\text{ONa}$
- d. $\text{C}_2\text{H}_5\text{ONa}$

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