

CHEM 102 - EXAM #2 QUESTIONS WITH COMPLETE SOLUTIONS

What is the name of the compound AlCl_3 ? - ANSWER-aluminum chloride

What is the name of the compound LiCl . - ANSWER-Lithium chloride

What is the name of the compound $\text{Mg}_3(\text{PO}_4)_2$? - ANSWER-magnesium phosphate

What is the name of the compound SO_3 ? - ANSWER-sulfur trioxide

Deduce the valence of aluminum in Al_2O_3 - ANSWER-3

Deduce the valence of calcium in CaO . - ANSWER-2

What are the vertical columns of the periodic chart called? - ANSWER-families

Which family of elements in the periodic table stands out most distinctly on the molar volume-atomic mass graph? - ANSWER-Group 1A

As a group, the 1A elements are the most chemically reactive of the periodic table. What conclusion follows most obviously from this fact and the molar volume curve? - ANSWER-large atoms are reactive

Calculate the amount of energy required to melt 500 g of ice at 0°C . ($\Delta H_{\text{fus}}=5.96 \text{ kJ/mole}$; ΔH_{fus} is the energy required to melt ice at a constant 0°C ; NOTE THE UNITS!) - ANSWER-39,600 cal

Calculate the amount of energy required to evaporate 500 g of water at 100°C . ($\Delta H_{\text{vap}}=40.5 \text{ kJ/mole}$; ΔH_{vap} is the energy to evaporate water; again, NOTE the units!) - ANSWER-269,000 cal

The melting point of lead bromide is 373 degrees Celsius. What is this in Kelvin? - ANSWER-646 K

Nitrogen, the major component of our atmosphere, boils at 96K. What is the Celsius value? - ANSWER--177°C

The average molecular kinetic energy is directly proportional to which quantity. - ANSWER-Kelvin temperature

The bond dissociation energy for nitrogen is 941 kJ/mol. What is this value in calories? - ANSWER-2.25x10⁵ cal/mol

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Unveiling the Energy of Verbal Art: An Emotional Sojourn through **Pkgacp Chem 102**

In some sort of inundated with screens and the cacophony of quick communication, the profound power and emotional resonance of verbal beauty usually disappear in to obscurity, eclipsed by the continuous onslaught of sound and distractions. Yet, located within the musical pages of **Pkgacp Chem 102**, a charming perform of literary beauty that pulses with natural feelings, lies an unforgettable journey waiting to be embarked upon. Penned by a virtuoso wordsmith, that interesting opus instructions visitors on an emotional odyssey, softly exposing the latent possible and profound affect embedded within the complicated internet of language. Within the heart-wrenching expanse with this evocative examination, we shall embark upon an introspective exploration of the book is key themes, dissect their fascinating publishing model, and immerse ourselves in the indelible impression it leaves upon the depths of readers souls.

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