CHEM 101 Formula & Data Sheet

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4	79 K 39.50	20 Ca 40.08	21 Sc 44.96	22 Ti 47.87	23 V 50.94	24 Cr 52:00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 NI 56.69	29 Cu 63.55	30 Zn 65.38	31 Ga 69.72	32 Ge 72.63	33 As 24.92	34 Se 78.96	35 84 79.90	36 Kr 83.60
5	37 855 85.47	38 54 8762	29 Y 88.95	40 2r 91.22	41 Nb 92.96	42 Mo 95.96	43 Tc (988)	84 Ru 1013	45 Rh 102.9	46 Pd 106.4	47 Ag 107.9	48 Cd 10.4	49 8n 114.8	50 5n ma.7	50 50 121.8	52 Te 1276	53 1 126.9	.54 Xe 131.3
6	55 Cs 132.9	5-6 Ba 137.3	57 La 138.9	72 HHF 179.5	73 To 180.9	74 W 103.0	75 Res 105.2	76 Os 190.2	77 br 19.2.2	79 Pt 1953	79 Au 197.0	80 Hg 200.6	81 TI 204.4	82 Pb 2022	83 88 209.0	84 Po (20%)	85 At (250)	86 Rn (222)
7	67 Fy 62230	Ro (2/24)	89 Ac (227)	104 Ref (2455)	Dib Chillia	106 50 (270	107 Bis (270)	100 His (277)	1019 MH (2270)	710 De (201)	Pig (2000)	110 Cm (295)	NB (2184)	204 (F) (21070)	Mic (2000)	196 Ew (2593)	117 -Tis (2-9-4)	00g

16-	Lanthanides	58 Ce 140.1	5/9 PY 140/9	60 Nd 144.2	64 Pm (945)	62 Sen 150-4	63 Ew 152.0	64 Gd 957.0	65 Tb 158.9	66 Dy 162.5	Ho Ho	6/8 Ex 167.3	Ten 160:0	70 Yb 173.1	71 Lu 175.0
7	Activides	90 Th 232.0	90 Pa (231)	92 U 238.0	93 Np (237)	.94 Pui (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (255)	99 ; Es (252)	100 Pen (257)	101 Mdd (21540)	No.2: No. (25/9)	103 Lr (262)

Formulas

$$E = hv = \frac{hc}{\lambda}$$
 $c = \lambda v$ $E_n = -R_H \left(\frac{Z^2}{n^2}\right)$ $\Delta E_n = -R_H Z^2 \left(\frac{1}{n_f^2} - \frac{1}{n_f^2}\right)$
 $FC = NVE - \text{(unshared e}^- + \frac{1}{2} \text{ shared e}^-\text{)}$ $BO = \frac{1}{2} \text{ (bonding e}^- - \text{antibonding e}^-\text{)}$

 $\Delta H^*_{rxn} \approx \Sigma \text{ BE(bonds broken)} - \Sigma \text{ BE (bonds formed)}$ Electrostatic force = $-k \left(\frac{q_1 q_2}{r^2}\right)$

$$(KE)_{avg} = \frac{3}{2} RT \qquad u_{ave} = \sqrt{\frac{3RT}{M}} \qquad P_i = x_i P_T \qquad x_i = \frac{n_i}{n_{ave}} = \frac{P_i}{P_{ave}}$$

$$PV = nRT \qquad (P + \frac{an^2}{V^2})(V - nb) = nRT$$

$$V(sphere) = \frac{4}{3} \pi r^3 \qquad V(cube) = \ell^3 \qquad BCC: \quad \ell = \frac{4 r}{\sqrt{3}} \qquad FCC: \quad \ell = \sqrt{8} r$$

$$(KE)_{tree} = \frac{1}{2}RT$$
 $u_{tree} = \sqrt{\frac{3Kt}{M}}$ $P_t = x_t P_T$ $x_t = \frac{3c_t}{P_{tree}} = \frac{2}{P_{tree}}$

$$V(subara) = \frac{4}{3} \pi r^3 \qquad V(suba) = t^3 \qquad BCC \quad t = \frac{4\pi}{3} \qquad ECC \quad t = 3$$

Constants and Conversion Factors

0.0 °C = 273.15 K	
1 cal = 4.184 J	
$1 \text{ eV} = 1.602 \times 10^{19} \text{ J} = 96.5 \text{ kJ mol}^{-1}$	
$1 \text{ nm} = 10^9 \text{ m}$	
$1 \text{ Å} = 10^{-10} \text{ m}$	
$1 \text{ pm} = 10^{-12} \text{ m}$	
	1 cal = 4.184 J 1 eV = 1.602×10^{-19} J = 96.5 kJ mol ⁻¹ 1 nm = 10^{-9} m 1 Å = 10^{-10} m

Pkgacp Chem 101

Michael Ash

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Handbook of Green Chemicals Michael Ash, 2004 More than 7000 trade name products and more than 2500 generic chemicals that can be used in formulations to meet envionmental concerns and government regulations This reference is designed to serve as an essential tool in the strategic decision making process of chemical selection when focusing on human and environmental safety factors Industries Covered Adhesives Refrigerants Water Treatment Plastics Rubber Surfactants Paints Coatings Food PharmaceuticalsCosmetics Petroleum Processing Metal Treatment TextilesThe chemicals and materials included are used in every aspect of the chemical industry. The reference is organized so that the reader can access the information based on the trade name chemical components functions and application areas green attributes manufacturer CAS number and EINECS ELINCS number It contains a unique cross reference that groups the trade name chemicals by one or more of these green chemical attributes Biodegradable Environmentally Safe Environmentally Friendly Halogen Free HAP s Free Low Global WarmingLow Ozone Depleting Nonozone Depleting Low Vapor Pressure Noncarcinogenic Non CFC Non HCFCNonhazardous Nontoxic Recyclable SARA Nonreportable SNAP Significant New Alternative Policy CompliantVOC Compliant Low VOC VOC Free Chemical Industries Week ,1954 Handbook of Paint and Coating Raw Materials: Trade name products ,1996 Multi-Agent Programming Helmut Werner, Gerhard Erker, 1989 Multi Agent Programming is an essential reference for anyone interested in the most up to date developments in MAS programming While previous research has focused on the development of formal and informal approaches to analyze and specify Multi Agent Systems this book focuses on the development of programming languages and tools which not only support MAS programming but also implement key concepts of MAS in a unified framework Part I describes approaches that rely on computational logic or process algebra Jason 3APL IMPACT and CLAIM SyMPA Part II presents languages and platforms that extend or are based on Java JADE Jadex and JACKTM Part III provides two significant industry specific applications The DEFACTO System for coordinating human agent teams for disaster response and the ARTIMIS rational dialogue agent technology Also featured are seven appendices for quick reference and comparison Soap, Cosmetics, Chemical Specialties ,1975 Plating and Surface Finishing ,1994 Helvetica Chimica Acta ,1987 Handbook of Preservatives Michael Ash, 2004 This handbook contains comprehensive information on more than 5000 trade names and generic chemicals and materials that are used in a broad range of formulations to prevent the contamination and decomposition of end products Product degradation can be caused by exposure to oxygen ozone bacteria molds yeast mildew and fungi The industries that depend on the proper selection of preserving chemicals and materials are diverse and include plastics elastomers construction paper pulp agriculture textiles paints and coatings pharmaceutical cosmetics food beverages This handbook contains comprehensive information on a variety of preservatives available from major chemical manufacturers and can expedite the material selection process for chemists formulators and purchasing agents by providing the answers to these questions Is the agent

capable of inhibiting the detrimental effects of oxygen ozone or microbes to the extent necessary Is the agent s overall physical and chemical attributes compatible with the product or system being protected Can the agent remain stable under storage conditions and for the application requirements Is its safety in production and handling acceptable Does its level of toxicity meet environmental regulations Does it meet cost requirements The Chemistry of Heterocyclic Compounds Druggists' Circular ,1922 **Drug and Chemical Markets** ,1927 American Druggists' Circular and ,1950 Official Directory Oregon. Office of Superintendent of Public Instruction, 1925 Chemical Gazette .1925 Chemical Industries ,1943 Some vols include Buyers guide The Chemical Trade Journal and Chemical Engineer ,1927 The Chemical Trade Journal and Chemical Engineer G Kelville Davis, 1927 Handbook of Fillers, Extenders, and **Drug Topics Price Book** ,1976 Diluents Michael Ash.2007 Food Engineering ,1968 Food Processing & Marketing ,1966-07

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