

# CER Practice

## CER Practice 1

**Question:** Does a chemical change occur if you mix sodium chloride solution and sucrose solution?

### Materials:

- sodium chloride solution (5 g of sodium chloride dissolved in 50 ml of water)
- sucrose solution (5 g of glucose dissolved in 50 ml of water)
- graduated cylinder
- balance
- clear glass beaker
- stirring rod
- thermometer
- red food coloring
- blue food coloring

### Procedure:

1. Place 5 g of sodium chloride into 50 ml of water and stir until it is dissolved.
2. Add red food coloring.
3. Place 5 g of sucrose into 50 ml of water and stir until it is dissolved.
4. Add blue food coloring.
5. Place a thermometer in the sodium chloride solution.
6. Pour the glucose solution into the sodium chloride solution.
7. Record your observations and the temperature.

### Data/Observations:

	Reactants		Products
	sodium chloride solution	sucrose solution	sodium chloride solution with sucrose solution
			clear, transparent
			ETC
			purple
			liquid

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## CER Practice 2

**Question:** Does a chemical change occur if you mix magnesium sulfate and sodium carbonate?

### Materials:

- magnesium sulfate solution (10 g of magnesium sulfate dissolved in 100 ml of water)
- sodium carbonate solution (5 g of sodium carbonate dissolved in 50 ml of water)
- test tube
- two droppers
- graduated cylinder
- balance
- clear plastic cup
- stirring rod
- thermometer

### Procedure:

1. Place 10 g of magnesium sulfate into 100 ml of water and stir until it is dissolved.
2. Place 5 g of sodium carbonate into 50 ml of water and stir until it is dissolved.
3. Place a thermometer into the magnesium sulfate solution and record the starting temperature.
4. Place the sodium carbonate solution into the magnesium sulfate solution.
5. Record your observations and the temperature.

### Data/Observations:

Appearance
Temperature
Color
State of Matter
Other

✓ **State your Claim**

✓ **Back up with Evidence**

✓ **Support with Reasoning**

# Chemical Reactions

# Pkgacp Cer Chemistry 123

**Jiyuan Zhang**



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## **Table of Contents Pkgacp Cer Chemistry 123**

1. Understanding the eBook Pkgacp Cer Chemistry 123
  - The Rise of Digital Reading Pkgacp Cer Chemistry 123
  - Advantages of eBooks Over Traditional Books
2. Identifying Pkgacp Cer Chemistry 123
  - Exploring Different Genres
  - Considering Fiction vs. Non-Fiction
  - Determining Your Reading Goals
3. Choosing the Right eBook Platform
  - Popular eBook Platforms
  - Features to Look for in an Pkgacp Cer Chemistry 123
  - User-Friendly Interface
4. Exploring eBook Recommendations from Pkgacp Cer Chemistry 123

- Personalized Recommendations
- Pkgacp Cer Chemistry 123 User Reviews and Ratings
- Pkgacp Cer Chemistry 123 and Bestseller Lists
- 5. Accessing Pkgacp Cer Chemistry 123 Free and Paid eBooks
  - Pkgacp Cer Chemistry 123 Public Domain eBooks
  - Pkgacp Cer Chemistry 123 eBook Subscription Services
  - Pkgacp Cer Chemistry 123 Budget-Friendly Options
- 6. Navigating Pkgacp Cer Chemistry 123 eBook Formats
  - ePub, PDF, MOBI, and More
  - Pkgacp Cer Chemistry 123 Compatibility with Devices
  - Pkgacp Cer Chemistry 123 Enhanced eBook Features
- 7. Enhancing Your Reading Experience
  - Adjustable Fonts and Text Sizes of Pkgacp Cer Chemistry 123
  - Highlighting and Note-Taking Pkgacp Cer Chemistry 123
  - Interactive Elements Pkgacp Cer Chemistry 123
- 8. Staying Engaged with Pkgacp Cer Chemistry 123
  - Joining Online Reading Communities
  - Participating in Virtual Book Clubs
  - Following Authors and Publishers Pkgacp Cer Chemistry 123
- 9. Balancing eBooks and Physical Books Pkgacp Cer Chemistry 123
  - Benefits of a Digital Library
  - Creating a Diverse Reading Collection Pkgacp Cer Chemistry 123
- 10. Overcoming Reading Challenges
  - Dealing with Digital Eye Strain
  - Minimizing Distractions
  - Managing Screen Time
- 11. Cultivating a Reading Routine Pkgacp Cer Chemistry 123
  - Setting Reading Goals Pkgacp Cer Chemistry 123
  - Carving Out Dedicated Reading Time
- 12. Sourcing Reliable Information of Pkgacp Cer Chemistry 123

- Fact-Checking eBook Content of Pkgacp Cer Chemistry 123
  - Distinguishing Credible Sources
13. Promoting Lifelong Learning
- Utilizing eBooks for Skill Development
  - Exploring Educational eBooks
14. Embracing eBook Trends
- Integration of Multimedia Elements
  - Interactive and Gamified eBooks

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