

CHEM 101 Formula & Data Sheet

1A (1)	2A (2)											3A (13)	4A (14)	5A (15)	6A (16)	7A (17)	8A (18)																
1 H 1.008		3 Li 6.941	4 Be 9.012											5 B 10.81	6 C 12.01	7 N 14.01	8 O 16.00	9 F 19.00	10 Ne 20.18														
2														11 Na 22.99	12 Mg 24.31											13 Al 26.98	14 Si 28.09	15 P 30.97	16 S 32.06	17 Cl 35.45	18 Ar 39.95		
3																																	
4																																	
5																																	
6																																	
7																																	

4

Be

9.012

Atomic number

Atomic symbol

Atomic mass (amu)

TRANSITION ELEMENTS																	
38 (38)	48 (48)	50 (50)	68 (68)	78 (78)	88 (88)			98 (98)	108 (108)	118 (118)	128 (128)	138 (138)	148 (148)	158 (158)	168 (168)	178 (178)	188 (188)
19 K 39.09	20 Ca 40.08	21 Sc 44.96	22 Ti 47.87	23 V 50.94	24 Cr 52.00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 Ni 58.69	29 Cu 63.55	30 Zn 65.38	31 Ga 69.72	32 Ge 72.63	33 As 74.92	34 Se 78.96	35 Br 79.90	36 Kr 83.80
37 Rb 85.47	38 Sr 87.62	39 Y 88.91	40 Zr 91.22	41 Nb 92.91	42 Mo 95.96	43 Tc (98)	44 Ru 101.1	45 Rh 102.9	46 Pd 106.4	47 Ag 107.9	48 Cd 112.4	49 In 114.8	50 Sn 118.7	51 Sb 121.8	52 Te 127.6	53 I 126.9	54 Xe 131.3
55 Cs 132.9	56 Ba 137.3	57 La 138.9	72 Hf 178.5	73 Ta 180.9	74 W 183.8	75 Re 186.2	76 Os 190.2	77 Ir 192.2	78 Pt 195.1	79 Au 197.0	80 Hg 200.6	81 Tl 204.4	82 Pb 207.2	83 Bi 209.0	84 Po (209)	85 At (210)	86 Rn (222)
87 Fr (223)	88 Ra (226)	89 Ac (227)	104 Rf (261)	105 Db (262)	106 Sg (266)	107 Bh (264)	108 Hs (277)	109 Mt (270)	110 Ds (285)	111 Rg (289)	112 Cn (285)	113 Nh (284)	114 Fl (289)	115 Mc (289)	116 Lv (293)	117 Ts (294)	118 Og (294)

6	Lanthanides	58 Ce 140.1	59 Pr 140.9	60 Nd 144.2	61 Pm (145)	62 Sm 150.4	63 Eu 152.0	64 Gd 157.3	65 Tb 158.9	66 Dy 162.5	67 Ho 164.9	68 Er 167.3	69 Tm 168.9	70 Yb 173.1	71 Lu 175.0
7	Actinides	90 Th 232.0	91 Pa (231)	92 U 238.0	93 Np (237)	94 Pu (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (251)	99 Es (252)	100 Fm (257)	101 Md (258)	102 No (259)	103 Lr (262)

Formulas

$$E = h\nu = \frac{hc}{\lambda} \quad c = \lambda \nu \quad E_n = -R_H \left(\frac{Z^2}{n^2} \right) \quad \Delta E_n = -R_H Z^2 \left(\frac{1}{n_f^2} - \frac{1}{n_i^2} \right)$$

$$FC = NVE - (\text{unshared } e^- + \frac{1}{2} \text{ shared } e^-) \quad BO = \frac{1}{2} (\text{bonding } e^- - \text{antibonding } e^-)$$

$$\Delta H^\circ_{rxn} \approx \sum BE(\text{bonds broken}) - \sum BE(\text{bonds formed}) \quad \text{Electrostatic force} = -k \left(\frac{q_1 q_2}{r} \right)$$

$$(KE)_{avg} = \frac{3}{2} RT \quad u_{rms} = \sqrt{\frac{3RT}{M}} \quad P_i = x_i P_T \quad x_i = \frac{n_i}{n_{tot}} = \frac{P_i}{P_{tot}}$$

$$PV = nRT \quad (P + \frac{an^2}{V^2})(V - nb) = nRT$$

$$V(\text{sphere}) = \frac{4}{3} \pi r^3 \quad V(\text{cube}) = \ell^3 \quad \text{BCC: } \ell = \frac{4r}{\sqrt{3}} \quad \text{FCC: } \ell = \sqrt{8} r$$

Constants and Conversion Factors

$N_A = 6.022 \times 10^{23} \text{ mol}^{-1}$	$1 \text{ atm} = 760 \text{ mmHg} = 760 \text{ torr} = 101.3 \text{ kPa}$
$h = 6.626 \times 10^{-34} \text{ J s}$	$0.0^\circ \text{C} = 273.15 \text{ K}$
$c = 2.998 \times 10^8 \text{ m s}^{-1}$	$1 \text{ cal} = 4.184 \text{ J}$
$R_H = 2.179 \times 10^{-18} \text{ J}$	$1 \text{ eV} = 1.602 \times 10^{-19} \text{ J} = 96.5 \text{ kJ mol}^{-1}$
$R = 0.08206 \text{ L atm mol}^{-1} \text{ K}^{-1}$	$1 \text{ nm} = 10^{-9} \text{ m}$
$= 0.08314 \text{ L bar mol}^{-1} \text{ K}^{-1}$	$1 \text{ \AA} = 10^{-10} \text{ m}$
$= 8.314 \text{ J mol}^{-1} \text{ K}^{-1}$	$1 \text{ pm} = 10^{-12} \text{ m}$

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Liotta



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**pkg Acp Cer Organic Chemistry Chem 3401* Liotta,2008

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