Chem 200 Exam 3 With Actual Solution (2025)!!

Electron Geometry using the VSEPR theory - Answer 2 electron groups-linear, 180 degrees

3 electron groups-trigonal planer, 120 degrees

4 electron groups- tetrahedral, 109.5 degrees

Molecular Geometry using the VSEPR - Answer 2 bonds-linear

2 bonds 1 lone pair-bent

3 bonds- trigonal planar

2 bonds 2 lone pairs- bent

3 bonds 1 lone pair- trigonal pyramidal

4 bonds- tetrahedral

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