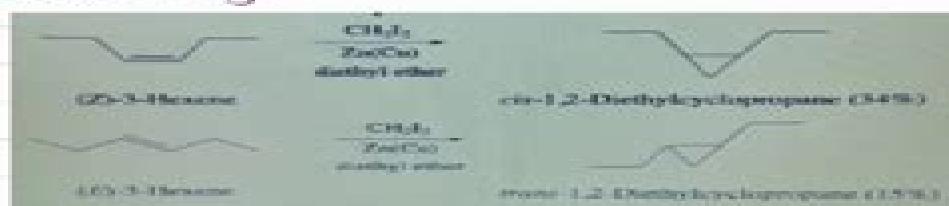


The Simmons-Smith Reaction

- Electron flow in one direction forges a C-C bond @ one carbon of $C=C$; e^- flow in the other direction forges C-C @ the other carbon of $C=C$.
- Organozinc reagent $(I_2n)CH_2I$ is both nucleophilic and electrophilic - Similar to a $C=C$ π bond.
- A cyclopropane product forms. Both bonds form simultaneously.



d-Block Elements

- The transition metals (d-block) occupy the center of the periodic table.
- They can form cations w/ a variety of charges & often sit @ the center of complexes w/ organic molecules @ the periphery.
- Transition-metal complexes are extremely important in modern chemical synthesis.
 - Their properties vary
 - Set of possible ligands are enormous
 - Have unique reactivity.

Transition-Metal Complexes

- A Lewis-acidic ($+n$) atom/ion surrounded by Lewis basic organic molecules (ligands). (definition)
- TMs are usually cationic + Lewis acidic in stable compounds.

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