Multiple-choice Questions, 1-14 (Canvas), are 2 points each and Questions 16-19 are 6 points each. (Total Points 52)

* For full credit on problems 15-18 show all of your work, including equations, calculations AND units*

- __1) Which of the following compounds will be more soluble in acidic solution than in pure water?
 - A) PbCl₂ B) FeBr₂
- C) Ca₂(PO₄)₂
- D) Nal
- E) None of the above will be more soluble in acidic solution.
- __2) Determine the molar solubility of Fe(OH)2 in pure water. K_{SD} for Fe(OH)2) = 4.87 × 10-17.
 - A) 2.44 × 10-17 M
 - B) 1.62 × 10-17 M
 - C) 4.03×10^{-9} M
 - D) 3.65 × 10 6 M
 - E) 2.30 × 10-6 M

__3) A solution containing CaCl2 is mixed with a solution of Li2C2O4 to form a solution that is 2.1×10^{-5} M in calcium ion and 4.75×10^{-5} M in oxalate ion. What will happen once these solutions are mixed? K_{20} (CaC2O4) = 2.3×10^{-9} .

- A) A precipitate will form since Q > Km for calcium oxalate.
- B) Nothing will happen since both calcium chloride and lithium oxalate are soluble compounds.
- C) Nothing will happen since calcium oxalate is extremely soluble.
- D) Nothing will happen since K_{SP} > Q for all possible precipitants.
- E) There is not enough information to determine.

The molar solubility of Ag2S is 1.26 × 10-16 M in pure water. Calculate the K_{sm} for Ag2S.

- A) $1.59 \times 10-32$
- B) 1.12 × 10-8
- C) 6.81 × 10-63
- D) 3.78×10^{-12}
- E) 8.00×10^{-48}

__5) A solution contains 3.8 × 10-2 M in Al3+ and 0.29 M in F at equilibrium. If the Kf for AlF63- is 7 × 10¹⁹, what is the concentration of the AlF63- ion at equilibrium?

- A) 1.52 × 1015 M
- B) $3.1 \times 10^{-22} \text{ M}$
- C) 9.1×10^{19} M
- D) 1.9×10^{-21} M
- E) 9.6 × 1015 M

_6) Which of the following processes have a △S > 0?

- A) 2 NH3(g) + CO2(g) \rightarrow NH2CONH2(aq) + H2O(l)
- B) lithium fluoride forms from its elements

C) 2 HBr(g) → H2(g) + Br2(l)

- D) sodium chloride dissolves in pure water
- E) All of the above processes have a ΔS > 0.

Pkg Acp Chemistry 115

Simardeep Kaur, Naseeb Singh, Gurjeet Singh, Rakesh Bhardwaj

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Rioux, Neidig, 2002-06-01 Pkg Acp-Chemistry 107 Lab Man GILLETTE, 2002-07-01 **Pkg Acp-Chemistry 105** Gillette, 2002-09-01 Integrative Multi-omics and Artificial Intelligence (AI)-driven Approaches for Superior Nutritional Quality and Stress Resilience in Crops Simardeep Kaur, Naseeb Singh, Gurjeet Singh, Rakesh Bhardwaj ,2025-10-08 The integration of multi omics and AI driven approaches represents a cutting edge frontier in agricultural research aiming to enhance the nutritional quality of crops Multi omics combines genomics transcriptomics proteomics metabolomics and other omics technologies to provide a comprehensive understanding of biological systems at multiple levels These technologies facilitate the identification of key genes pathways and metabolites that contribute to nutritional traits and stress resilience in crops Artificial Intelligence AI further amplifies this potential by analyzing complex omics data predicting outcomes and quiding precise genetic modifications By utilizing AI researchers can expedite the breeding process optimize metabolic pathways and tailor crops to meet specific nutritional goals This integrative approach promises to revolutionize crop improvement strategies addressing global challenges related to food security malnutrition and agricultural sustainability The convergence of multi omics and AI not only accelerates the development of nutrient rich crops but also enhances their adaptability to environmental stresses ensuring a more resilient and nutritious food supply Pkg Acp-Chemistry 230 Manivre, 2002-09-01 Highlights of POG 2019 - Plant Oxygen Group Conference Christian Lindermayr, Joerg Durner, Ann Cuypers, Jörg-Peter Schnitzler, Krystyna Oracz, 2021-03-09 **Pkg Acp Chemistry 109** Neidig, 2002-09-01 **Pkg Acp-Chemistry 1211** Markow, 2002-09-01 Pkg Acp-Gen Chemistry Ii Lab Griswold, 2002-06-01 Pkg Popular Photography ,1991-01 **Acp-Organic Chemistry** Manivre, 2002-06-01 Pkg Acp-Chem 170 Markow, 2002-10-01

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