

CHE 120: BASIC CHEMISTRY II

PRACTICE QUESTIONS

QUESTION ONE

a) i. State the Avogadro's

ii. Write short notes on the order of reaction and rate constants

iii. In an investigation of the recombination of X atoms to give X gas phase in the presence of argon, that is the reaction $X + X + Ar$ following data were obtained.

With the concentration of argon fixed at $1.0 \times 10^{-2} \text{ mol dm}^{-3}$

Concentration [X] (mol dm^{-3})	Initial rate [dX_2/dt] ($\text{mol dm}^{-3}\text{s}^{-1}$)
1.0×10^{-3}	8.7×10^{-4}
2.0×10^{-3}	3.48×10^{-3}
4.0×10^{-3}	1.39×10^{-2}

With the concentration of X fixed at $1.0 \times 10^{-3} \text{ mol dm}^{-3}$

Concentration [X] (mol dm^{-3})	Initial rate [dX_2/dt] ($\text{mol dm}^{-3}\text{s}^{-1}$)
1.0×10^{-3}	8.7×10^{-4}
5.0×10^{-3}	4.35×10^{-3}
1.0×10^{-2}	8.69×10^{-3}

Find the order of reaction with respect to [X] and [Ar], hence determine the overall velocity constant for the formation of X

QUESTION TWO

a) At certain temperature, equilibrium constant for the reaction below is 5.0.



ii the reaction started with initial partial pressures of CO and steam as 1 atmosphere and 10 atmospheres, calculate the partial pressure of all the gases when the equilibrium state is reached.

b) A 2.5 M solution of a weak monobasic acid has the same pH as 0.017M HCl. Assuming that the HCl is completely ionized.

i. Calculate the pH of these solutions

ii. Determine the degree of ionization of the weak acid in its 2.5M solution.

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