



**University of the Western Cape**  
**Chemistry Department**

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**CHEMISTRY 116**  
**Test 2**

April 2017

MARKS: 20

TIME: 1 1/2 Hour

- ANSWER ALL QUESTIONS.
  - USE THE ANSWER SHEET PROVIDED.
  - A PERIODIC TABLE IS PROVIDED.
- 

12. Which of the following is a balanced chemical equation?

- A.  $\text{H}_2\text{O}_2 \rightarrow \text{H}_2\text{O} + \text{O}_2$
- B.  $2 \text{Fe}_2\text{O}_3 + 3 \text{C} \rightarrow 4\text{Fe} + 3\text{CO}_2$
- C.  $\text{SO}_2 + \text{O}_2 + 2\text{H}_2\text{O} \rightarrow 4\text{H}_2\text{SO}_4$
- D.  $2 \text{Mg} + \text{HCl} \rightarrow \text{MgCl}_2 + \text{H}_2$

13. The combustion of propane ( $\text{C}_3\text{H}_8$ ) in the presence of excess oxygen yields  $\text{CO}_2$  and  $\text{H}_2\text{O}$ :



When 2.5 mol of  $\text{O}_2$  are consumed in their reaction, \_\_\_\_\_ mol of  $\text{CO}_2$  are produced.

- A) 1.5
- B) 2.5
- C) 5.0
- D) 6.0
- E) 3.0

Consider the following chemical reaction for problems, 14, 15, 16, 17, 18



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