



Chem 110B Practice Final Exam

Part A. Please answer all questions thoroughly.

1. What energy equivalent appears above the arrow in the following calculated reaction?
 $\text{p-NO}_2\text{Cl} \rightarrow \text{p-NO}_2\text{Cl} + \text{H}_2\text{O}$

eq 1 eq 2 eq 3 eq 4 eq 5 _____

2. What energy equivalent appears below the arrow?

eq 6 eq 7 eq 8 eq 9 eq 10 _____

3. What energy equivalent is below the arrow in the following reaction?

eq 11 or 12 eq 13 or 14 eq 15 or 16 eq 17 or 18 _____

4. What is the following reaction the product of the following reaction?

eq 19 or 20 eq 21 or 22 eq 23 or 24 _____

5. What volume of 0.1 M NaOH is needed to neutralize the reaction of 0.1 M HCl and 0.1 M H_2SO_4 ?

eq 25 or 26 eq 27 or 28 eq 29 or 30 eq 31 or 32 _____

6. Complete the following table by supplying the chemical formula of the compound.

Source	Formula
	PbCl_2
Calcium phosphate	
	Na_2SO_4
	Calcium

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