

CHEM 102 - EXAM #2 QUESTIONS WITH COMPLETE SOLUTIONS

What is the name of the compound AlCl_3 ? - ANSWER-aluminum chloride

What is the name of the compound LiCl . - ANSWER-Lithium chloride

What is the name of the compound $\text{Mg}_3(\text{PO}_4)_2$? - ANSWER-magnesium phosphate

What is the name of the compound SO_3 ? - ANSWER-sulfur trioxide

Deduce the valence of aluminum in Al_2O_3 - ANSWER-3

Deduce the valence of calcium in CaO . - ANSWER-2

What are the vertical columns of the periodic chart called? - ANSWER-families

Which family of elements in the periodic table stands out most distinctly on the molar volume-atomic mass graph? - ANSWER-Group 1A

As a group, the 1A elements are the most chemically reactive of the periodic table. What conclusion follows most obviously from this fact and the molar volume curve? - ANSWER-large atoms are reactive

Calculate the amount of energy required to melt 500 g of ice at 0°C . ($\Delta H_{\text{fus}}=5.96 \text{ kJ/mole}$; ΔH_{fus} is the energy required to melt ice at a constant 0°C ; NOTE THE UNITS!) - ANSWER-39,600 cal

Calculate the amount of energy required to evaporate 500 g of water at 100°C . ($\Delta H_{\text{vap}}=40.5 \text{ kJ/mole}$; ΔH_{vap} is the energy to evaporate water; again, NOTE the units!) - ANSWER-269,000 cal

The melting point of lead bromide is 373 degrees Celsius. What is this in Kelvin? - ANSWER-646 K

Nitrogen, the major component of our atmosphere, boils at 96K. What is the Celsius value? - ANSWER--177°C

The average molecular kinetic energy is directly proportional to which quantity. - ANSWER-Kelvin temperature

The bond dissociation energy for nitrogen is 941 kJ/mol. What is this value in calories? - ANSWER-2.25x10⁵ cal/mol

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