

3. Do you expect propanol to be miscible with water? Why or why not?

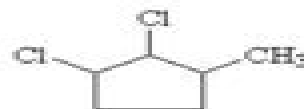
Miscible. C_3H_7OH is small and has dipole-dipole and hydrogen bonds like H_2O .

4. Which of the following chemical formulas could be a cycloalkane?

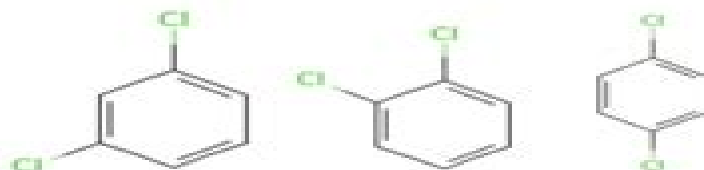
- a. C_5H_{10}
- b. C_2H_4
- c. C_6H_{10}
- d. C_4H_{10}

5. What is the IUPAC name for this molecule?

- a. 1-methyl-2,3-dichlorocyclopentane
- b. 1,2-dichloro-3-methylcyclopentane
- c. 1,2-dichloro-3-methylcyclobut-1-ane
- d. 3-methyl-1,2-dichlorocyclopent-1-ane
- e. 1,2-dichloro-3-methyl-1-pentane



6. Name the following molecules. Which, if any, are nonpolar?



meta-dichlorobenzene, *ortho*-dichlorobenzene, *para*-dichlorobenzene

The only nonpolar one is *para*-dichlorobenzene because the Cl's cancel out. These molecules are planar.

7. According to VSEPR, what geometry does ethyne have?

- a. Linear
- b. Bent
- c. Trigonal Planar
- d. Tetrahedral

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Emil J. Slowinski



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