#### CHEM 101 Formula & Data Sheet

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2	3 E8 6-944	8e 9.052		.012		Abornic mass (amu) 5 6 7 8 C N 93.81 12.01 14.01									6 0 16.00	9 F 19:00	Ne 20.18	
*	H No 22.99	Mg 24.31	30 (3)	400	540 (50	- TRAI	78 (7)	COLUMN COLUMN	88 - 69	(20)	100	210	AJ 26.98	54 51 28 09	15 P 30.97	96 5 32.06	17 CI 3/5.4/5	18 Ar 39.95
4	79 K 39.50	20 Ca 40.08	21 Sc 44.96	22 Ti 47.87	23 V 50.94	24 Cr 52:00	25 Mn 54.94	26 Fe 55.85	27 Co 58.93	28 NI 56.69	29 Cu 63.55	30 Zn 65.38	31 Ga 69.72	32 Ge 72.63	33 As 24.92	34 Se 78.96	35 84 79.90	36 Kr 83.60
5	37 855 85.47	38 54 8762	29 Y 88.95	40 2r 91.22	41 Nb 92.96	42 Mo 95.96	43 Tc (988)	84 Ru 1013	45 Rh 102.9	46 Pd 106.4	47 Ag 107.9	48 Cd 10.4	49 8n 114.8	50 5n ma.7	50 50 121.8	52 Te 1276	53 1 126.9	.54 Xe 131.3
6	55 Cs 132.9	5-6 Ba 137.3	57 La 138.9	72 HHF 179.5	73 To 180.9	74 W 103.0	75 Res 105.2	76 Os 190.2	77 br 19.2.2	79 Pt 1953	79 Au 197.0	80 Hg 200.6	81 TI 204.4	82 Pb 2022	83 88 209.0	84 Po (20%)	85 At (250)	86 Rn (222)
7	67 Fy 62230	Ro (2/24)	89 Ac (227)	104 Ref (2455)	Dib Chillia	106 50 (270	107 Bis (270)	100 His (277)	1019 MH (2270)	710 De (201)	Pig (2000)	110 Cm (295)	NB (2184)	204 (F) (21070)	Mic (2000)	196 Ew (2593)	117 -Tis (2-9-4)	00g

16-	Lanthanides	58 Ce 140.1	5/9 PY 140/9	60 Nd 144.2	64 Pm (945)	62 Sen 150-4	63 Ew 152.0	64 Gd 957.0	65 Tb 158.9	66 Dy 162.5	Ho Ho	6/8 Ex 167.3	Ten 160:0	70 Yb 173.1	71 Lu 175.0
7	Activides	90 Th 232.0	90 Pa (231)	92 U 238.0	93 Np (237)	.94 Pui (244)	95 Am (243)	96 Cm (247)	97 Bk (247)	98 Cf (255)	99 ; Es (252)	100 Pen (257)	101 Mdd (21540)	No.2: No. (25/9)	103 Lr (262)

#### **Formulas**

$$E = hv = \frac{hc}{\lambda}$$
  $c = \lambda v$   $E_n = -R_H \left(\frac{Z^2}{n^2}\right)$   $\Delta E_n = -R_H Z^2 \left(\frac{1}{n_f^2} - \frac{1}{n_f^2}\right)$   
 $FC = NVE - \text{(unshared e}^- + \frac{1}{2} \text{ shared e}^-\text{)}$   $BO = \frac{1}{2} \text{ (bonding e}^- - \text{antibonding e}^-\text{)}$ 

 $\Delta H^*_{rxn} \approx \Sigma \text{ BE(bonds broken)} - \Sigma \text{ BE (bonds formed)}$  Electrostatic force =  $-k \left(\frac{q_1 q_2}{r^2}\right)$ 

$$(KE)_{avg} = \frac{3}{2} RT \qquad u_{ave} = \sqrt{\frac{3RT}{M}} \qquad P_i = x_i P_T \qquad x_i = \frac{n_i}{n_{ave}} = \frac{P_i}{P_{ave}}$$
 
$$PV = nRT \qquad (P + \frac{an^2}{V^2})(V - nb) = nRT$$
 
$$V(sphere) = \frac{4}{3} \pi r^3 \qquad V(cube) = \ell^3 \qquad BCC: \quad \ell = \frac{4 r}{\sqrt{3}} \qquad FCC: \quad \ell = \sqrt{8} r$$

$$(KE)_{tree} = \frac{1}{2}RT$$
  $u_{tree} = \sqrt{\frac{3Kt}{M}}$   $P_t = x_t P_T$   $x_t = \frac{3c_t}{P_{tree}} = \frac{2}{P_{tree}}$ 

$$V(subara) = \frac{4}{3} \pi r^3 \qquad V(suba) = t^3 \qquad BCC \quad t = \frac{4\pi}{3} \qquad ECC \quad t = 3$$

#### Constants and Conversion Factors

0.0 °C = 273.15 K	
1 cal = 4.184 J	
$1 \text{ eV} = 1.602 \times 10^{19} \text{ J} = 96.5 \text{ kJ mol}^{-1}$	
$1 \text{ nm} = 10^9 \text{ m}$	
$1 \text{ Å} = 10^{-10} \text{ m}$	
$1 \text{ pm} = 10^{-12} \text{ m}$	
	1 cal = 4.184 J 1 eV = $1.602 \times 10^{-19}$ J = $96.5$ kJ mol <sup>-1</sup> 1 nm = $10^{-9}$ m 1 Å = $10^{-10}$ m

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Jessica J Manson

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