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 Lab Partner: Katelin H.
 Chem 201, Section 18
 9/19/23

CHEM201: EXPERIMENT 3
Using the pH Meter to Measure K_a and K_b

PROCEDURE:

Refer to my pre lab report page 04 of the lab notebook to see an outline of the procedure. No changes were made in this lab.

EXPERIMENTAL DATA:

Table 1: pH Values

Sample	0.10 M Solution	0.001 M Solution
NH_3	11.40	10.40
NH_4^+	6.90	7.11

CALCULATED RESULTS

Table 2: Calculated Results

Solution	$[H_3O^+]$	$[OH^-]$	$[NH_4^+]$	$[NH_3]$	K_a & K_b	% ionization
0.1M NH_3	3.98×10^{-12}	2.54×10^{-3}	3.98×10^{-12}	0.1	1.58×10^{-11}	$1.26 \times 10^{-6} \%$
0.001M NH_3	3.98×10^{-12}	2.54×10^{-3}	3.98×10^{-12}	0.001	3.98×10^{-8}	$7.76 \times 10^{-5} \%$
0.1M NH_4^+	1.26×10^{-7}	8.02×10^{-8}	0.1	1.26×10^{-7}	1.59×10^{-11}	$3.98 \times 10^{-6} \%$
0.001M NH_4^+	7.76×10^{-8}	1.30×10^{-7}	0.001	7.76×10^{-8}	6.02×10^{-12}	$3.98 \times 10^{-6} \%$

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