# 113L

#### **CHEMISTRY LAB**

#### ACTIVITY 5

COMPOUND IDENTIFICATION OF UNKNOWN DRUG SAMPLES USING PHYSICAL PROPERTIES

#### **PHYSICAL PROPERTIES**

- 1. QUALITY ASSURANCE
- Detection of physical changes
  - Presence of all contaminants and instability are a sign of poor quality
- 2. STABILITY ASSESSMENT
- Monitor and assess stability
  - Emulsion (separation of oil and water)
    is a detection of instability
- 9. EARLY DETECTION OF CONTAMINATION
- Reveal the presence of contaminants/impurities
  - The formation or presence of molds
  - and/or precipitations is an early detection of contamination
- 4. BATCH TO BATCH CONSISTENCY
- Same au ali ty within batches
  - Botch no. series of numbers.
  - Lot no. portion of the batch

#### SOLUBILITY

#### PRINCIPLE

- Like dissolves like meaning polar solutes dissolve in polar solvents
- Solubility product (Ksp)
  - Solubility product constant
  - Indication to determine if a sample is contaminated but it should not be the sole basis to determine it
  - Ksp values are found in pharmacopeias
  - Sx = 2.00 -> Ksp = 1.6
    - It means that Sx is contaminated because it does not match the one in the pharmacopeia
  - What is reflected in the USP are pure substances

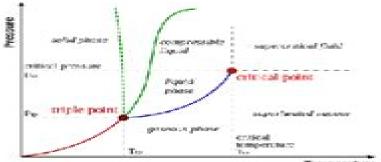
#### MIDTERMS

#### PURPOSE

- Formulation Development
- Properly formulate drugs and maximize intended effects by incorporating excipients
- 2. Drug Delivery Systems
- Desired drug release profiles
- 3. Biografiability
- Solubility and absorption

#### MELTING POINT RANGE

"The temperature from the point at which solid first begins to liquify to the point at which the entire sample is liquid."



Termonie reductor

#### PRINCIPLE

- 1. Eutectic Point (Solid to Liquid)
- Lowest temperature where phase transformation is observed
- Phase Transition (Solid to Liquid)
- Very narrow temperature range

#### Kinetic Energy and IMFA

 Inversely proportional (Increased KE, decreased IMFA)

#### Triple Point

Where solid, liquid and gas will exist simultaneously

#### Critical Point

. Difficult for liquid to be converted into gas state

#### Supercritical Fluid

 Impossible to convert liquid into its gas state (High pressure)

# **Pkgacp Chemistry 113l Lab Manual**

Antony C. Wilbraham, Dennis D. Staley, Michael S. Matta, Edward L. Waterman

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