

Claim: Unknown solution #5 is potassium chromate, or K_2CrO_4 .

Evidence: Unknown solution #5 turned purple in the flame test, just like potassium turned purple. It also changed a darker color and got moderately cloudy from the reactions of HCl and $BaCl_2$, just like chromate did.

Reasoning: Because the reactions from the HCl and $BaCl_2$ and the flame test was the same as potassium and chromate, unknown solution #5 has to be K_2CrO_4 , or Potassium Chromate.

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Neidig



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