



Use the exact values you enter to make later calculations.

Experiment 7: Reduction-Oxidation Titrations

 $2 \ \mathrm{KMnO_4(aq)} + 5 \ (\mathrm{COOH})_2(aq) + 3 \ \mathrm{H_2SO_4(aq)} \rightarrow 10 \ \mathrm{CO_2(g)} + 2 \ \mathrm{MnSO_4(s)} + \mathrm{K_2SO_4(aq)} + 8 \ \mathrm{H_2O(I)}$

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