CHEM 1405 Practice Exam # 4 (Ch 8-10)

Section # Name MULTIPLE CHOICE. Choose the one alternative that best completes the statement or answers the question. 1) Which one of the following is characteristic of a base? 1) _____ A) has a sour taste. B) is insoluble in water C) produces H₃O* in water D) turns blue litmus red E) has a slippery, soapy feel 2) Which of the following is correctly identified? 2) A) HCl, weak acid B) Ca(OH)2, weak base C) NaOH, strong base D) NHa, strong acid E) H2CO3, strong acid 3) The conjugate base of H₂O is _____ 3) _____ A) OHT B) O2-C) H₃O* D) H₂O E) H₂O has no conjugate base. 4) Which of the following is the strongest acid? B) NH4" A) HaPOs C) H₂CO₃ D) HCI E) NaOH. 5) When a reaction is at equilibrium, ___ A) the reaction is no longer reversible B) the products and reactants have the same energy content C) no more reactants are converted to products D) the forward and reverse reactions occur at the same rate E) all reaction stops 6) What is the pH of a solution with [OH⁻] = 1 x 10⁻⁴ M? 6) _____ A3 - 10.0B) 10.0 C) 4.0: D0 = 4.0E) 1.0×10^{-10}

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