

Point Counterpoint 8 Cases For Composition

Yijin Wang

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web the limiting reactant or limiting reagent is the reactant that gets consumed first in a chemical reaction and therefore limits how much product can be formed as we saw in example 1 there are many different ways to determine the limiting reactant but they all involve using mole ratios from the balanced chemical equation

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web the key to recognizing which reactant is the limiting reactant is to determine the reactant that yields the least amount of product is the limiting reactant it doesn t matter which product is used for the calculation as long as the same product is