PHASE DIAGRAMS FOR CERAMISTS: SUPPLEMENT NO. 1

BY HOWARD F. McMURDIR AND F. P. HALL*

Постиновычеством:

The original compilation of phase diagrams was made by r. P. Hall and Herbert Insley and was published as the October, 1903, issue of The Journal of The American Cerowic Society. These authors then followed with a supplement, published as the April, 1908, issue of the same Jearnal, and with another complete compilation which was published as Part II of the November, 1947, issue of this Jearnal.

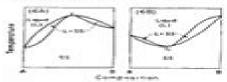
The following compilation is the first supplement to the November, 1947, issue. It includes corrections of errors? and omitted diagrams as well as those published since the 1947 compilation was printed.

It is to be understood that temperatures are given in degrees centigrade and the compositions are given in weight percentage unless otherwise stated on the individual diagram.

ERRATA, NOTES, AND COMMENTS

Page 6 (ii) and Page 7, Fig. 6. (Corrected Fig. 6. Note on caption to Fig. 6 and first paragraph of (ii).)

In a two-component condensed system (constant pressure), P + F = C + 1 and F = 3 - P. In



For. 6. (corrected).—Supersedes Fig. 6, page 7.

Binary systems with a single solid solution: (A), system with a maximum melting point at C, which is not a compound; (B), system with a minimum melting point at C, which is not a suspen-

Handbuch der Metallphysik. Edited by Georg Masing. Vol. II, Die heterogenen Gleichgewichte, by Rudolf Vogel, p. 241. Akademische Verlagsgesellschaft, Berlin, 1937.

tem to be invariant (F = 0), three phases must exist in equilibrium. Such a condition, however, can never exist in a solid solution series as shown. in Fig. 6, because there are never TOTAL CHARGE a Branch B WILLIAMS phases present, i.e., solid solution and liquid solution. The system can at no point become invariant and the equilibrium curves. must be continuous. in contrast to a "true" in wariant.

order for the sys-

point (entectic or peritectic) which is a point of discontinuity on the curve.

Such points designating maxima and minima, including the melting points of congruently melting compounds, at which two phases become identical in composition, are known as indifferent points. Alex-

ander Findlay and A. N. Campbell, Phase Rule and Its Applications, 8th ed., pp. 114 and 121. Dover Publications, New York, 1945. xv + 313 pp. + 163 figs.

Page 13, (d), second paragraph, lines 6-9, should read:

. . . ; from d to e, the tangent intersects an extension of AB-A, hence A crystallizes again and AB is partially resorbed; at e, A is once more partially redissolved while AB and C crystallize out, and the final products of crystallization are A, AB, and C.

Page 23, Fig. 41.—System BaO-SiO₂.

In the reference by P. Eskola, the date (122) should read (1922).

Page 24, Fig. 47.—System CuO-SiOs.

The compound 3CaO·SiO₂ is stable only in the temperature range of about 1300° to 1900°C. Above 1900°, it decomposes into α-2CaO·SiO₂ and CaO; below 1300°C., it decomposes into β-2CaO·SiO₂ and CaO. E. T. Carlson, J. Remarch Natl. Bur. Standards, 7 [5] 893–902 (1931): RP 381.

Page 25, Fig. 51.—System Fe-Fe₇O₄.

More recent work in the system Fe-Fe₂O₂ is shown in Fig. 510, page 156.

Page 26, Fig. 52—System Fe-Fe₂O₄.

More recent work in the system Fe-Fe₂O₂ is shown in Fig. 510, page 156.

Page 29, Fig. 69.—System MgO-ZrO₁.

More recent work on zirconia (ZrO₂) suggests that the temperature-stability relation shown on this diagram is doubtful.

Page 34, Fig. 92.—System SiO₁-ZrO₁.

A revised and corrected diagram in the system SiO₂-ZrO₃ is shown in Fig. 513, page 157.

^{*} The authors are, respectively, chief, Constitution and Microstructure Section, National Bureau of Standards, Washington, D. C., and Director of Research, Pass and Seymour, Inc., Syracuse, New York.

[†] The authors wish to acknowledge the assistance of many friends in pointing out the errors that appeared in the 1947 compilation of phase diagrams.

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Phase Diagrams in Advanced Ceramics Allen M. Alper, 1995-02-08 The investigation of multi component complex systems composed of oxides nitrides and carbides has intensified in the last few years Phase Diagrams in Advanced Ceramics reviews some of the recent advances in the understanding of these composite systems providing insight into how phase diagrams can be utilized in the fabrication of whiskers and ceramic matrix whisker reinforced ceramics Phase relations and sintering information is reviewed for transparent polycrystalline oxides Phase diagrams are discussed to predict alkali oxide corrosion of alumino silicate references Understanding the development manufacture and use of complex multi component ceramic materials composed of silicon nitride metal oxides nitride carbide systems Development and use of whisker and whisker reinforced ceramics composed of materials such as alumina silicon nitride silicon carbide and directly solidified eutectic ceramics Application of phase diagrams to the production of advanced composites such as alumina matrix zirconium diboride and titanium hafnium zirconium carbides and borides Phase chemistry in the development of transparent poly crystal and oxides including yttria alumina and magnesium aluminate Improvements concerning the knowledge of complex multi component materials composed of oxides nitrides and carbides and knowledge of how to fabricate composite materials containing whiskers and ceramic hosts New developments in making transparent ceramic materials Phase Diagrams for Ceramists Robert S. Roth, Jennifer R. Dennis, American Ceramic Society, Howard F. McMurdie, 1987 User Evaluation of "Phase Diagrams for Ceramists" And Implications for Related Data and Research Programs (Classic Reprint) R. S. Roth, 2017-12-14 Excerpt from User Evaluation of Phase Diagrams for Ceramists And Implications for Related Data and Research Programs Phase diagrams play an important role in the development of ceramic materials Phase equilibria data are essential in meeting the expanding needs for refractories electronic components non crystalline solids and various other applications of interest to ceramic scientists and engineers The chemical information generally summarized as a phase equilibrium diagram is extremely useful in many industrial processes However the necessary research involved in obtaining such information is often very costly. The average binary phase diagram can be estimated to take about one man year and a

ternary diagram may take five times that effort To avoid duplication it is advantageous to have all existing diagrams compiled and distributed for easy access As described below the American Ceramic Society long ago recognized the importance of phase diagrams to the field of ceramic materials and the importance of having this type of information available in evaluated and summarized form Beginning in 1933 the Society has worked contin uously with the National Bureau of Standards to provide a sequence of reports under the title Phase Diagrams for Ceramists giving evaluated data from the literature on selected ceramic systems The Bureau has supplemented these data with measurements in its own laboratories on a limited number of systems chosen because the data were needed for specific reasons but were not available or were of doubtful accuracy Other laboratories such as the Geophysical Laboratory of the Carnegie Institute and the Pennsylvania State University have supported Phase Diagrams for Ceramists through evaluations by their staffs and or through experimentalwork The overall responsibility and general editorship have however always been assumed by the National Bureau of Standards Several valuable contributions to compilation evaluation and publication of phase diagrams of ceramic systems have been made by other groups and published independently from time to time as described below These have generally been one time only activities with a limited specialized scope Phase Diagrams for Ceramists is the only continuing program of general scope in the Western World With the recent addition of an International Board of Contributing Editors it appears very likely to be even more central and unique in the future About the Publisher Forgotten Books publishes hundreds of thousands of rare and classic books Find more at www forgottenbooks com This book is a reproduction of an important historical work Forgotten Books uses state of the art technology to digitally reconstruct the work preserving the original format whilst repairing imperfections present in the aged copy In rare cases an imperfection in the original such as a blemish or missing page may be replicated in our edition We do however repair the vast majority of imperfections successfully any imperfections that remain are intentionally left to preserve the state of such historical works Phase Equilibria Diagrams Anna E. McHale, 1993 Phase Equilibria Diagrams ,1992 Phase Equilibria Diagrams ,1995 **Phase Equilibria Diagrams** Gerald B. Stringfellow, 1992 Phase Diagrams for Ceramists ,1981

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