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web characteristics physical acids would look solid liquid or in the form of gas it would also have a sour taste bases would feel slippery and solid in nature except for ammonia

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web log H 10.4 05.8 91 $\times 10^{-5}$ mol dm^{-3} log pK_a pH 3.75 4.05 0.30 10.0 30.0 50 and $HCOOH$ 0.1 so $HCOO^-$ 0.200 mol dm^{-3} amount vol x

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web only the better candidates could write successive ionisation equations in part c while part d about the pH of equimolar strong and weak acid solutions was poorly answered

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web a substance may be assigned to one of our four conceivable categories it may be an acid or a base but in addition it may be both an acid and a base or it may be neither an acid nor

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web to detect the presence of acids and bases Michael adds a few drops of phenolphthalein indicator to two colorless solutions a and b he observes the following changes in the

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web Nov 5 2023 most of the acids are solid it has a sour taste they produce hydrogen after reacting with active metals like

zinc magnesium aluminium or iron acids have a low ph

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web section 2 ph of strong acids number of protons released monoprotic acid acid that releases one h ion per molecule e g hcl hydrochloric acid hno₃ nitric acid ch

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web jan 10 2021 s14 3 21 strong electrolytes are 100 ionized and as long as the component ions are neither weak acids nor weak bases the ionic species present result

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web problem 1 which of the following are not acids ch₃cooh co₂ hno₂ hcooh cc₄ c h₃ c o o h c o₂ h n o₂ h c o o h c c l₄ shazia naz numerade educator 02 23

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web in grade 11 learners were introduced to a lot of the concepts that are expanded on in this chapter these include acid and base models and definitions conjugate acid base

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web which is the conjugate base in the following reaction multiple choice question 13 which of the following compounds is the strongest acid multiple choice question 14 which

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web 0 0 change x x x

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web 2 at what ph or range of ph values is a solution considered strongly acidic mildly acidic neutral mildly basic and strongly basic provide a brief explanation

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web 11 why is hcl a strong acid and hclo a weak acid 12 why are hcl and hclo 4 both strong acids 13 for each of the reactions below classify the reactants as an acid or a

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web answer i lemon and orange ii tomatoes and guava iii sour milk curd iv tamarind question 18 a student detected the ph of four unknown solution a b c and d as