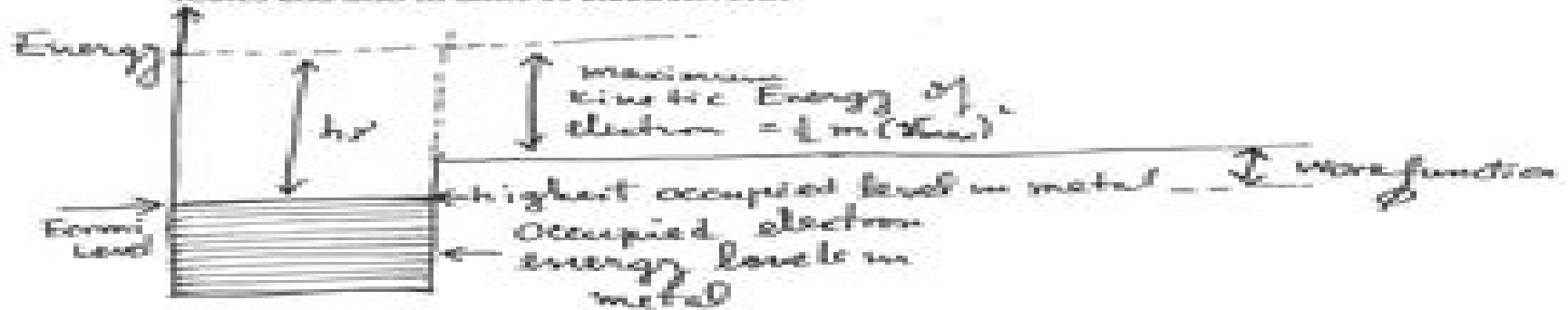


## Workshop 2

### Question 1

The longest wavelength of light that will induce emission of electrons from potassium is 564 nm. Calculate the work function of potassium in units of Joules and also in units of electronvolts.



The minimum energy required to release an electron from the Fermi level is known as the work function  $\phi$  and can be expressed as

$$h\nu = T + \phi \quad (1)$$

where  $T$  is the maximum kinetic energy of the emitted electrons.

At the critical frequency  $\nu_0$ , photoelectrons are emitted with zero kinetic energy and then equation (1) becomes:  $\lambda\nu_0 = \phi$

$$c = \lambda_0 \nu_0 \quad \Rightarrow \quad \phi = \frac{hc}{\lambda_0}$$

$$\phi = \frac{(6.626 \times 10^{-34} \text{ Js}) \times (2.998 \times 10^8 \text{ m/s})}{564 \times 10^{-9} \text{ m}} = 3.52 \times 10^{-19} \text{ J}$$

Let's convert in units of  $(\text{eV} \sim 1.602 \times 10^{-19} \text{ C})$

$$1 \text{ eV} = (1.602 \times 10^{-19} \text{ C}) \text{ V} = 1.602 \times 10^{-19} \text{ J}$$

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