

1. (4 points) Arrange these compounds by boiling point. 1 = lowest boiling point.



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2. (4 points) Arrange these compounds by water solubility. 1 = most water-soluble.



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3. (6 points) Draw the bond-line structure of the following molecules.



4. (6 points) Draw the condensed structure of the following molecules.



5. (6 points) Identify the 2 functional group present in each molecule. Designate 1°, 2°, 3°, terminal/internal, if appropriate.



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6. (4 points) Use the molecule below to fill in the blanks.



# of 1° carbons \_\_\_\_\_

# of 2° carbons \_\_\_\_\_

# of 1° hydrogens \_\_\_\_\_

# of 2° hydrogens \_\_\_\_\_

7. (4 points) What major IR band would help you determine the difference for each pair of molecules? Name what type of bond and the position (in  $\text{cm}^{-1}$ ) of where you would expect to see the difference.



# Sg And Drill Problems Organic Chemistry

**L Towner**



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